Synthesis of 4‑Fluoromethylsydnones and their Participation in Alkyne Cycloaddition Reactions

Robert S. Foster,[†] Harry Adams,[†] Harald Jakobi,[‡] and Joseph P. A. Harrity^{*,†}

† Department of Chemistry, University of Sheffield, Sheffield S3 7HF, U.K.

‡ Research-Weed Control Chemistry, Bayer CropScience, AG 65926 Frankfurt am Main, Germany

S Supporting Information

[AB](#page-14-0)STRACT: [We report t](#page-14-0)he synthesis and some structural studies of 4-trifluoromethyl, 4 difluoromethyl-, and 4-monofluoromethylsydnones. All but the latter compounds are stable and represent effective precursors to a range of pyrazoles after cycloaddition reactions with alkynes. The cycloadditions are generally highly regioselective and provide 5-fluoromethylpyrazole products, although we have observed that Bn-substituted sydnones can provide an unexpected alkyne insertion mode that generates the 3-fluoromethyl isomer.

■ INTRODUCTION

The prevalence of fluorinated aromatic and heterocyclic compounds in agrochemicals and small molecule therapeutics has led to a significant interest in developing new and more convenient synthetic procedures for the preparation of these molecules. As a specific subclass, pyrazoles bearing fluorocarbon substituents are becoming increasingly prevalent as synthetic targets and building blocks within the fine chemicals sector.¹ Several bioactive fluorinated pyrazoles have emerged in recent years and have shown activity in pharmaceutical² and agroc[he](#page-14-0)mical screens.³ Of the more established commercial fluorinated pyrazoles, particularly noteworthy examples i[nc](#page-14-0)lude nonsteroidal anti-infla[m](#page-14-0)matory drugs of the coxib class such as celecoxib and deracoxib,⁴ as well as the herbicide fluazolate⁵ (Figure 1).

The $[3 + 2]$ cycloaddition of nitrilimines and alkynes and the cyclocondensation of a hydrazine with 1,3-diketones or α , β unsaturated carbonyl compounds represent the most common strategies of pyrazole ring synthesis. However, these approaches can suffer from the formation of regioisomeric mixtures with respect to substituents incorporated at the pyrazole 3- and 5 positions.⁶ In the context of trifluoromethylpyrazoles, these compounds are typically synthesized by the cyclocondensation approach using 1,1,1-trifluoromethyl-1,3-diketones, as this method exploits the ready availability of trifluoroacetic acid derived precursors.⁷ Such reactions also very often provide mixtures of the 3- and 5-trifluoromethyl products, although good regiocontrol [fo](#page-14-0)r the 3 -CF₃ pyrazoles can be achieved by use of fluorinated solvents.⁸ Moreover, the efficient synthesis of $5-CF_3$ -substituted pyrazoles is further hampered by the fact that the intermediate 5-trifluor[om](#page-14-0)ethyl-5-hydroxypyrazolines undergo rather slow dehydration.⁹ As an alternative to the cyclocondenstaion route, the cycloaddition of alkynes with sydnones represents a conven[ie](#page-14-0)nt approach for the regioselective synthesis of these azoles.^{10,11} In a related study, Meazza reported the synthesis of 3- and 4-trifluoromethylpyrazoles through cycloadditions with trifl[uoro](#page-14-0)methylacetylenes.¹² Moreover, diazomethane functions as an effective precursor to stannylated 4-fluoro- and 4-trifluoromethylated py[raz](#page-14-0)oles.¹³ However, employing cycloaddition techniques to provide the analogous 5-trifluoromethylpyrazoles is rather under-develop[ed](#page-14-0) (Scheme 1). Recent studies in our laboratory have endeavored to develop the scope of sydnone functionalization and alkyne cycloaddi[tio](#page-1-0)n chemistry, with the goal of establishing this area as enabling chemistry for pyrazole synthesis.¹⁴ We envisaged that this chemistry could provide a convenient and general solution to the regiocontrolled synthesi[s](#page-14-0) of various 5 fluoromethyl-substituted pyrazoles via the corresponding 4 fluoromethylsydnones.¹⁵ Our studies toward this end are delineated herein.

■ RESULTS AND DISCUSSION

Sydnones are typically prepared in two steps by nitrosation of N-substituted amino acids followed by cyclodehydration of the resulting nitrosamines. Therefore, we began our investigations by developing a scalable route to the requisite 4-trifluoromethylsydnones via their respective trifluoromethylalanines.

Received: February 21, 2013 Published: April 2, 2013

Scheme 1. Synthetic Approaches to Trifluoromethylpyrazoles

Scheme 2 shows the route employed to access compounds 5a,b. The aza-Wittig reaction of iminophosphoranes 1a,b and

Scheme 2. Synthesis of 4-Trifluoromethylsydnones a

methyl trifluoromethylpyruvate furnished imines 2a,b,¹⁶ which were reduced to the corresponding amino esters 3a,b using zinc metal.¹⁷ Ester hydrolysis proved to be challenging; sa[po](#page-14-0)nification provided a complex mixture whereas hydrolysis under acid cataly[sis](#page-14-0) proved to be capricious. Ultimately, however, we found that heating the amino esters at reflux with lithium iodide in ethyl acetate for 4 h^{18} consistently delivered the amino acids 4a,b in high yield. Finally, the desired 4-trifluoromethylsydnones were generate[d](#page-14-0) by the well-established nitrosation− cyclodehydration sequence, thereby allowing gram quantities of 5a and 5b to be produced in good overall yield.

Our efforts to further extend this chemistry to produce other trifluoromethylsydnones highlighted some limitations with this approach. Specifically, our attempts to reduce the PNP-imine 6 with zinc metal resulted in mixtures of the reduction products.

Recourse to alternative reducing agents such as trichlorosilane only gave trace amounts of the desired amino ester.

Overall therefore, with a view to implementing a general approach to trifluoromethylsydnones, the route outlined in Scheme 2 raised some serious limitations. In addition to the potential for functional group incompatibility highlighted in Scheme 3, the synthetic sequence is lengthy, linear, and

Scheme 3. Uncontrolled Reduction of N-PNP Imine 6^a

a PNP: 4-nitrophenyl.

incorporates the N-substituent at the start of the process. We envisaged that a more convenient strategy would be to develop a late-stage trifluoromethylation of sydnones, as this would provide a more direct means to incorporate a range of Nsubstituents. In this regard, we reported that the Suzuki crosscoupling of 4-bromosydnones provided a convenient method of incorporating aryl-substituents.^{14b} We envisaged that an analogous strategy could prove to be an effective method for generating 4-trifluoromethyl syd[non](#page-14-0)es.

Copper-promoted trifluoromethylations using Ruppert's reagent are one of the most widely used techniques for the direct incorporation of a CF_3 group,¹⁹ although the use of organoboron-based coupling partners is also well established.²⁰ In the event, however, successful triflu[oro](#page-14-0)methylation of 7 was achieved by the Cu-promoted addition of methyl fluorosulf[o](#page-14-0)nyldifluoroacetate 21 in DMF at 80 °C. The scope of the trifluoromethylation was examined, and the results are summarized in S[ch](#page-14-0)eme 4. The reaction of 4-iodo-N-phenyl-

Scheme 4. Trifluoromethylation of 4-Iodosydnones

sydnone 7 ($R = Ph$) provided trifluoromethylsydnone 5a in 79% yield. The presence of an electron-donating PMP group was also highly efficient, giving 5b in similar yield. However, the electron-withdrawing PNP group required a much longer reaction time but provided the previously inaccessible compound 5c in acceptable yield. Nonaromatic groups were also tolerated, delivering the methyl and benzyl protected sydnones 5d and 5e. Overall, therefore, the direct trifluor-

omethylation described in Scheme 4 complements and improves upon the stepwise procedure described earlier (Scheme 2) and in our preliminary rep[or](#page-1-0)t of this work¹⁵ and offers a practical approach to a selection of fluorous sydnone analogues.

Having developed effective routes to CF_3 -substituted sydnones, we next turned our attention to their cycloaddition reaction with alkynes to form pyrazoles. We were particularly interested in establishing the effect of the trifluoromethyl group on cycloaddition regioselectivity, and our results are summarized in Table 1. Alkyne cycloaddition reactions with N-Ph

sydnone 5a were generally complete within 24 h when conducted at 180 °C in a sealed vessel. We were pleased to find that the cycloaddition of 5a with a range of terminal alkynes proceeded in good to excellent yields and with almost complete regiocontrol in each case (entries 1−8). Indeed, the selectivity of formation of 11 is notable; cycloadditions of 4- Me- and 4-Prⁱ-substituted sydnones with 2-ethynylpyridine proceed with lower levels of regiocontrol $($6:1$)^{14f} suggesting$ that the CF_3 group can also enhance cycloaddition

regioselectivity in some cases. Moreover, disubstituted alkynes also participated in the reaction, but with generally lower levels of regiocontrol (entries 9−11). These general trends were continued with N-arylsydnones 5b and 5c, although cycloadditions of PNP-sydnone 5c needed only 8 h, in accord with the documented increased reactivity of these mesoionic compounds (entries $12-16$).^{14c} Finally, the cycloaddition of N-Me-sydnone 5d also provided the corresponding pyrazoles in excellent yield and regiocontr[ol](#page-14-0) (entries 17−20). However, in extending this method to the cycloaddition with 5-chloropent-1-yne, we were surprised to recover the bicyclic 3 trifluoromethyl pyrazole 29. In this case, we believe that pyrazole 28 is formed initially but then undergoes a cyclization to provide 29 upon demethylation (Scheme 5).

The study highlighted in Table 1 suggested that the chemistry would have limited use in the direct synthesis of tetrasubstituted pyrazoles, as the observed regioselectivities were found to be rather low (entries 9 and 10). Related work in our laboratories has highlighted the potential of alkynylboronates in the synthesis of these compounds, 14 as they typically afford high selectivites for the 4-borylated isomer. We therefore decided to investigate the cycloaddition of t[hes](#page-14-0)e substrates, and our results are summarized in Table 2. In the event, the reactions required heating at 140 °C over a period of 24−72 h; however, the corresponding products 30−34 were generated in good yield and with useful levels of regiocontrol. Notably, the terminal alkyne exhibited the opposite regiochemical insertion mode and follow the trend whereby the sterically more

a Product isolated as an inseparable mixture of regioisomers. o-DCB: 1,2-dichlorobenzene.

a Reaction complete within 8 h. PMP: 4-methoxyphenyl. PNP: 4 nitrophenyl. ^bProduct isolated as an inseparable mixture of isomers.

demanding group is placed adjacent to the nitrogen atom, in line with our previous observations.^{14c}

Representative functionalization reactions confirmed the potential of the pyrazole boronic [es](#page-14-0)ters for the divergent synthesis of 5 -CF₃ pyrazoles (Scheme 6). Both 3- and 4-

Scheme 6. Functionalization of 5-Trifluoromethylpyrazole Boronic Esters

borylated compounds underwent Suzuki coupling in good yields. Of particular note is the regioselective synthesis of 16a, which was produced as a 1:1 mixture by reaction of 5a with 1 phenyl-1-hexyne (see Table 1, entry 9). Similarly, both 3- and 4-borylated 30b and 32a underwent oxidation with hydrogen peroxide at ambient tempera[tu](#page-2-0)re to provide 35 and 36 in good yields. The product pyrazoles offer the opportunity to access the bioactive pyrazole phenyl ether series. 22

A limitation of the sydnone cycloaddition route to pyrazoles is the requirement for a substituent to be [p](#page-14-0)resent at nitrogen. However, we envisaged that PMP-substituted sydnone 5b offered the opportunity to generate pyrazoles bearing a free N-H after oxidative cleavage using ceric ammonium nitrate $(CAN)²³$ To ensure full conversion a large excess of CAN was required²⁴ which not only gave the expected pyrazole 37 in 47% yi[eld](#page-14-0) but also 13% of the nitrated product 38, which appears [to](#page-14-0) be deactivated toward oxidation (Scheme 7).

The low yield observed in the CAN deprotection of 19 prompted us to explore other systems that could offer more efficient deprotection strategies. In this respect, we decided to

Scheme 7. CAN Deprotection of N-PMP-Protected 5- Trifluoromethylpyrazole 19

examine the cycloaddition of 4-trifluoromethyl-N-benzylsydnone 5e, with a view to studying Bn hydrogenolysis of the product pyrazoles. To our surprise, however, the reaction of 5e with both phenylacetylene and 1-hexyne yielded a mixture of the expected 5 -CF₃-substituted pyrazoles 39a and 40a, together with the corresponding 3 -CF₃ isomers 39b and $40b^{25}$ (Table 3). We were able to improve the selectivity to favor regioisomer

Table 3. Synthesis of 3- and 5-Trifluoromethyl-Nbenzylpyrazoles

 $a^a48%$ conversion. b^b The values in parentheses refer to the equivalents of alkyne used.

a by lowering the reaction temperature (entries 3 and 4) or using a large excess of alkyne (entry 5), but in each case a small amount of the 3-trifluoromethyl isomer remained.

The observation that sydnones can incorporate alkynes into the 4,5-position of a pyrazole under thermal conditions is unprecedented to the best of our knowledge. In order to explain this curious result, we decided to first explore the potential for Bn-migration in the pyrazole products. As outlined in Scheme 8, subjecting samples of 39a,b containing different

Scheme 8. Probing Product Isomerization

ratios of regioisomers to heating under the reaction conditions resulted in recovery of the pyrazoles with no detectable changes in regioisomer ratios. We speculated that sydnone 5e could function as an electrophilic source of Bn-cation, thereby promoting product isomerization by a benzylation−debenzylation mechanism along similar lines to the process highlighted in Scheme 5. Again, however, heating mixtures of 39a,b in the presence of sydnone 5e resulted in no detectable changes in pyrazole [is](#page-2-0)omer ratios.

The control experiments outlined in Scheme 8 suggested that the 3 -CF₃ isomer did not originate via pyrazole isomerization, and we began to consider pathways under which the 3 trifluoromethyl isomer could originate from the sydnone. In this regard, 41 has been proposed to undergo a disrotary electrocyclic ring closure under photolytic conditions to give a bicyclic intermediate 43 which evolves $CO₂$ to produce a nonisolable nitrile imine 44. ²⁶ Alkyne cycloaddition of 44 with

The Journal of Organic Chemistry Article 30 and 200 an

dimethyl acetylenedicarboxylate (DMAD) gave pyrazole 42b in 67% yield. In contrast, the thermally promoted reaction of sydnone 41 with DMAD provides isomeric pyrazole 42a (Scheme 9).

Scheme 9. Photolytic and Thermal Reactions of Sydnones with DMAD

In order to test whether the formation of 39b was occurring via a visible light promoted nitrile imine formation, the cycloaddition with phenyl acetylene was conducted simultaneously in the presence and absence of light (Scheme 10). However, both reactions yielded approximately the same ratio of isomers under these conditions.

Scheme 10. Visible Light Dependence on the Cycloaddition of Sydnone 5e

In order to establish the effect of reaction temperature on product distribution, we exploited the enhanced reactivity of DMAD in sydnone cycloadditions so that 5e could be reacted with DMAD at 120 and 180 °C. At the lower temperature, the reaction produced $5-CF_3$ pyrazole 45a in 54% yield and essentially as a single regioisomer. In contrast, conducting the reaction at 180 °C produced a mixture of 45a and 45b in a ratio of 63:37 (Scheme 11).

Scheme 11. Comparison of Thermal Cycloaddition Reactions of Sydnone 5e

Taken together, these results suggest that the benzylsydnone 5e undergoes transformation to a new intermediate at elevated temperatures and that this undergoes cycloaddition to provide a 3 -CF₃ pyrazole product. We propose that a Bn migration takes place to form an intermediate 1,2,3-oxadiazolidin-5-one 46 that undergoes decarboxylation to generate nitrile imine 47 and that this dipolar intermediate is the reactive species in the $3-CF_3$ pyrazole-forming reaction. Indeed, Tanaka has demonstrated that a closely related intermediate trifluoroacetonitrile imine 49 undergoes cycloaddition with phenylacetylene to give 3 -CF₃pyrazole 50 (Scheme 12).²⁷ Studies aimed at providing

Scheme 12. Proposed Mec[han](#page-15-0)ism of Formation of 3-CF₃ Pyrazoles

experimental evidence for intermediates 46 and 47 and to explore other such migration reactions in 3,4-disubstituted pyrazoles are ongoing and will be reported in due course.

From a synthetic standpoint, our initial goal was to undertake cycloadditions of 5e so that we could carry out a subsequent debenzylation to furnish N-unsubstituted pyrazoles. In this regard, it is unimportant that we observe compound mixtures in this particular case as both isomers converge to a single pyrazole upon hydrogenolysis. For example, compounds 39a,b provided pyrazole 37 in 75% yield (Scheme 13).²⁸

Scheme 13. Catalytic Hydrogenolysis of Bn-Pro[te](#page-15-0)cted 3- and 5-Trifluoromethylpyrazoles

Having established a robust method for the introduction of the trifluoromethyl group into pyrazoles, we wondered if the concept could be extended to include monofluoromethyl and difluoromethyl substituents. While monofluoromethyl pyrazoles are rather rare, difluoromethyl-substituted analogues can be prepared via hydrazine condensation reactions with the respective difluoromethyl-1,3-diketones. As with the trifluoromethyl analogues, the 3-fluoromethyl isomer predominates.²⁹ Therefore, as an extension to the synthesis and cycloadditions of trifluoromethyl sydnones, we envisioned that the known [4](#page-15-0) formyl- and 4-hydroxymethylsydnones³⁰ could be transformed into the corresponding fluoromethylsydnones using deoxofluor. These sydnones could then be utilize[d a](#page-15-0)s novel precursors to diverse 5-fluoromethyl- and 5-difluoromethylpyrazoles.

Although the direct synthesis of 4-formyl-N-phenylsydnone is reported,³⁰ we had difficulties reproducing the published methods. However, we were able to generate the 4-ethyl ester 52 via lithi[atio](#page-15-0)n of N-phenylsydnone 51, followed by addition to ethyl chloroformate. Sydnone 52 was then reduced to the alcohol 53 in excellent yield by stirring with lithium borohydride overnight, and this alcohol could subsequently be oxidized to the 4-formyl sydnone 54 using $MnO₂$ (Scheme 14).

Scheme 14. Synthesis of Sydnones 53 and 54

Deoxofluorination of the hydroxymethyl sydnone 53 was highly efficient, providing 4-fluoromethyl-N-phenylsydnone 55 in 80% yield after stirring at ambient temperature for 2 h. The same reaction with 4-formyl-N-phenylsydnone 54 required significantly more forceful conditions (5 equiv of $[(CH₃OCH₂CH₂)₂NSF₃]$ deoxofluor, 48 h), but 4-difluoromethyl-N-phenylsydnone 56 was isolated in 81% yield (Scheme 15).

Scheme 15. Synthesis of Fluorinated Sydnones by Deoxofluorination

The cycloaddition of difluoromethylsydnone 56 with a selection of alkynes was undertaken, and the results are summarized in Table 4. The reaction was found to be generally very efficient and provided the corresponding 5-difluorome-

a Product isolated as an inseparable mixture of regioisomers. o-DCB: 1,2-dichlorobenzene.

thylpyrazoles in high yield with good regioselectivities. Notably, the sterically less demanding difluoromethyl group provides lower selectivities in the case of 2-ethynylpyridine and ethyl propiolate, in comparison to the trifluoromethyl analogue (compare entries 2 and 5 in Table 4 with entries 4 and 20 in Table 1).

We next turned our attention to the alkyne cycloaddition reacti[on](#page-2-0)s of 4-fluoromethyl-N-phenylsydnone 55. To our disappointment, the reaction of phenylacetylene resulted in decomposition. The situation could not be improved by reducing the temperature to 120 °C. We next opted to exploit the reactivity of DMAD and were pleased to find that cycloaddition could be achieved by conducting the reaction at 100 °C. However, as illustrated in Scheme 16, we discovered that the resulting pyrazole was actually the 5-hydroxymethyl analogue 63.

We decided to explore the susceptibility of sydnone 55 to hydrolysis by exposing this compound to deuteromethanol, allowing the reaction progress to be conveniently monitored by ¹H NMR spectroscopy. At room temperature, full conversion to sydnone 64 occurred over several days but this time was reduced to 44 h if the reaction mixture was heated at reflux (Scheme 17).

As we were unable to productively use fluoromethylsydnone 55 in our cycloaddition reactions, we synthesized 1,3-diphenyl-5-hydroxymethylpyrazole 65 from 4-hydroxymethyl-N-phenylsydnone 53 in 72% yield. The subsequent reaction with deoxofluor appeared to proceed to full conversion within 2 h .³¹ Unfortunately, however, upon workup either by quenching with $NAHCO_{3(aq)}$ or submitting directly to flash chromatography [on](#page-15-0) silica gel, a significant proportion of the desired fluorinated pyrazole 66 was found to have hydrolyzed to pyrazole 65, showing the susceptibility of this compound to hydrolysis (Scheme 18).

Finally, the 4-fluoromethylated sydnones 5a and 56 were crystallin[e,](#page-6-0) and we therefore decided to analyze these compounds by X-ray diffraction; the structures of both compounds are shown in Figure 2. The crystal structure of difluoromethyl sydnone 56 shows an alignment of the C−H and C−O bonds. The conformatio[n m](#page-6-0)ay simply reflect reduced A(1,3) strain, or favorable C−O and C−F bond dipole alignments (or a combination of both). Curiously however, the trifluoromethyl-substituted sydnone 5a highlights a similar alignment in the solid phase resulting in a close contact between the oxygen and fluorine atoms.³

Scheme 18. Attempted Synthesis of 5-Fluoromethylpyrazole 66

Figure 2. X-ray crystal structures of sydnones 5a and 56.

■ CONCLUSION

In conclusion, we have demonstrated that 4-trifluoromethylsydnones can be prepared by ring synthesis and ring functionalization approaches. These compounds function as valuable precursors to 5-trifluoromethylpyrazoles via highly regioselective cycloaddition reactions, although the N-Bn sydnone 5e delivers an unexpected alkyne insertion reaction to provide both 5- and 3- CF_3 pyrazoles. The CF_3 -pyrazole boronic esters can be further fucntionalized through established and widely utilized organoboron chemistry, demonstrated by the synthesis of key intermediates for biologically active compounds. This chemistry has been extended toward the synthesis of 4-fluoromethyl- and 4-difluoromethylsydnones. While there are inherent stability issues regarding the 4 fluoromethyl compounds, 4-difluoromethyl-N-phenylsydnone 56 proved to be a valuable precursor for the preparation of a range of 5-difluoromethylpyrazoles.

EXPERIMENTAL SECTION

Synthesis of Triphenylphosphine Phenylimide 1a. To a stirring solution of azidobenzene (15.86 g, 133.12 mmol) in ether (200 mL) was slowly added a solution of triphenylphosphine (34.92 g, 133.12 mmol) in ether (200 mL). The mixture was stirred for 2 h, and then the volatiles were removed in vacuo. The crude material was purified by recrystallization from ethanol to yield tetraphenylphosphine imide 1a as a yellow solid (37.25 g, 79%): mp 134 − 135 °C (lit.³³ mp 135−136 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.82−7.76 (m, 6H), 7.57−7.53 (m, 3H), 7.50−7.45 (m, 6H), 7.07−7.01 (m, 2H), 6.85[−](#page-15-0)6.81 (m, 2H), 6.71−6.65 (m, 1H); 13C NMR (100.6 MHz, CDCl₃) δ 151.1 (d, J = 2.5 Hz), 132.6 (d, J = 9.5 Hz), 131.7 (d, J = 2.5 Hz), 131.4 (d, J = 99.0 Hz), 128.7 (d, J = 12.0 Hz), 128.6, 123.5 (d, J = 17.5 Hz), 117.3; ³¹P NMR (101.1 MHz, CDCl₃) δ 2.89; HRMS m/z $[MH]^{+}$ calcd for $C_{24}H_{21}NP$ 354.1412, found 354.1411.

Synthesis of Triphenylphosphine (p-Methoxyphenyl)imide 1b. To a stirring solution of 1-azido-4-methoxybenzene (18.40 g, 123.37 mmol) in ether (200 mL) was slowly added a solution of triphenylphosphine (32.36 g, 123.37 mmol) in ether (200 mL). The mixture was stirred for 2 h, and then the volatiles were removed in vacuo. The crude material was purified by recrystallization from ethanol to yield triphenylphosphine (p-methoxyphenyl)imide 1b as a yellow solid (41.51 g, 88%): mp 116−118 °C (lit.³⁴ mp 117−118 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.82–7.78 (m, 6H), 7.54–7.50 (m, 3H), 7.47−7.42 (m, 6H), 6.83−6.81 (m, 2H), [6.6](#page-15-0)8−6.66 (m, 2H), 3.70 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 152.0, 144.5 (d, J = 2.5 Hz), 132.7 (d, $J = 9.5$ Hz), 131.6 (d, $J = 2.5$ Hz), 131.5 (d, $J = 99.0$ Hz), 128.6 (d, J = 12.0 Hz), 123.9 (d, J = 16.5 Hz), 114.3, 55.6; ³¹P NMR (101.1 MHz, CDCl₃) δ 2.49; HRMS m/z [MH]⁺ calcd for C₂₅H₂₃NOP 384.1517, found 384.1512.

Synthesis of (Z)-Methyl 3,3,3-Trifluoro-2-(phenylimino) **propanoate 2a.**¹⁶ To a stirring solution of methyl trifluoropyruvate (11.70 g, 75.00 mmol) in toluene (200 mL) was added solid tetraphenylphos[phin](#page-14-0)e imide 1a (26.50 g, 75.00 mmol) in one portion. The mixture was stirred for 2 h before the volatiles were removed in vacuo. The crude material was suspended in petroleum ether, filtered, and washed with more petroleum ether before the volatiles were removed in vacuo. This procedure was repeated until the precipitation of triphenylphosphine oxide had ceased to yield (Z)-methyl 3,3,3 trifluoro-2-(phenylimino)propanoate 2a as a yellow oil (15.95 g, 92%).

Note: Compound 2a could alternatively be purified by flash chromatography on silica gel (eluting with 10% EtOAc in petroleum ether): ¹ H NMR (400 MHz, CDCl3) δ 7.43−7.39 (m, 2H), 7.31−7.27 (m, 1H), 7.00−6.97 (m, 2H), 3.73 (s, 3H); 13C NMR (100.6 MHz, CDCl₃) δ 159.9, 148.5 (q, J = 37.0 Hz), 146.4, 129.2, 127.3, 119.2, 118.2 (q, J = 278.5), 52.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –69.8; FTIR 2930 (w), 2856 (w), 1749 (s), 1326 (m), 1263 (s), 1190 (m), 1155 (s), 1037 (s) cm⁻¹; HRMS m/z [MH] ⁺ calcd for C₁₀H₉F₃NO₂ 232.0585, found 232.0591.

Synthesis of (Z)-Methyl 3,3,3-Trifluoro-2-(para-methoxy**phenylimino)propanoate 2b.** 35 To a stirring solution of methyl trifluoropyruvate (2.84 g, 16.02 mmol) in toluene (25 mL) was added solid triphenylphosphine (p-me[tho](#page-15-0)xyphenyl)imide 1b (6.14 g, 18.17 mmol) in one portion. The mixture was stirred for 2 h before the volatiles were removed in vacuo. The crude material was suspended in petroleum ether, filtered, and washed with more petroleum ether before the volatiles were removed in vacuo. This procedure was repeated until the precipitation of triphenylphosphine oxide had ceased to yield (Z)-methyl 3,3,3-trifluoro-2-(p-methoxyphenylimino) propanoate 2b as an orange oil (3.77 g, 96%).

Note: Compound 1b could alternatively be purified by flash chromatography on silica gel (eluting with 10% EtOAc in petroleum ether): ¹H NMR (400 MHz, CDCl₃) δ 7.03 (d₁ J = 9.0 Hz, 2H), 6.92 $(d, J = 9.0 \text{ Hz}, 2\text{H})$, 3.83 (s, 3H), 3.79 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 160.9, 159.6, 146.1 (q, J = 37.0 Hz), 138.9, 122.4, 118.4 (q, J = 278.0 Hz), 114.5, 55.4, 52.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −69.6; FTIR 3462 (br), 2967 (w), 2845 (w), 1748 (s), 1508 (s), 1256 (s), 1162 (s), 1035 (s), 844 (m) cm[−]¹ ; HRMS m/ z [MH]⁺ calcd for $C_{11}H_{11}F_3NO_3$ 262.0691, found 262.0680.

Synthesis of Methyl 3,3,3-Trifluoro-2-(phenylamino) **propanoate 3a.**³⁶ To a stirring suspension of zinc dust (8.26 g) 126.31 mmol) in AcOH (75 mL) was added (Z)-methyl 3,3,3 trifluoro-2-(phen[ylim](#page-15-0)ino)propanoate 2a (14.60 g, 63.16 mmol). The mixture was stirred at ambient temperature for 2 h, filtered, and neutralized with NaHCO₃. The solution was extracted with EtOAc (3) \times 50 mL), washed with water (25 mL), and dried over MgSO₄ before the volatiles were removed to yield methyl 3,3,3-trifluoro-2- (phenylamino)propanoate 3a as a colorless oil (11.49 g, 78%): ¹H NMR (400 MHz, CDCl₃) δ 7.29−7.25 (m, 2H), 6.92−6.98 (m, 1H), 6.77−6.75 (m, 2H), 4.70−4.59 (m, 2H), 3.89 (s, 3H); 13C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 166.8, 145.1, 129.5, 123.3 (q, J = 283.5), 120.1, 114.2, 59.6 (q, J = 31.5 Hz), 53.6; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -72.7 (d, J = 6.0 Hz); FTIR 3391 (br), 3038 (w), 2965 (w), 2853 (w), 1756 (s), 1606 (s), 1514 (s), 1441 (m), 1304 (s), 1218 (s), 1183 (s)

cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₀H₁₁F₃NO₂ 234.0742, found 234.0744.

Synthesis of Methyl 3,3,3-Trifluoro-2-(p-methoxyphenylamino)propanoate 3b. To a stirring suspension of zinc dust (1.82 g, 27.81 mmol) in AcOH (15 mL) was added (Z)-methyl 3,3,3 trifluoro-2-(p-methoxyphenylimino)propanoate 2b (3.63 g, 13.91 mmol). The mixture was stirred at ambient temperature for 2 h, filtered, and neutralized with NaHCO₃. The solution was extracted with EtOAc $(3 \times 25 \text{ mL})$, washed with water (25 mL) , and dried over MgSO4 before the volatiles were removed in vacuo to yield methyl 3,3,3-trifluoro-2-(p-methoxyphenylamino)propanoate 3b as a yellow oil (2.45 g, 67%): ¹H NMR (400 MHz, CDCl₃) δ 6.83 (d, J = 9.0 Hz, 2H), 6.73 (d, J = 9.0 Hz, 2H), 4.57−4.50 (quint, J = 7.0 Hz, 1H), 4.39 $(d, J = 7.5 \text{ Hz}, 1H)$, 3.85 (s, 3H), 3.77 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 167.1, 154.0, 139.0, 123.4 (q, J = 283.0), 116.2, 115.0, 61.0 $(q, J = 31.0 \text{ Hz})$, 55.6, 53.5; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –72.7 $(d, J = 6.5 \text{ Hz})$; FTIR 3374 (br), 2955 (w), 3836 (w), 1758 (s), 1519 (s), 1245 (s), 1126 (s), 1032 (m), 823 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{11}H_{13}F_3NO_3$ 264.0848, found 264.0853.

Synthesis of 3,3,3-Trifluoro-2-(phenylamino)propanoic Acid 4a. To a stirring solution of methyl 3,3,3-trifluoro-2-(phenylamino) propanoate 3a (3.33 g, 14.28 mmol) in EtOAc (60 mL), shielded from light, was added lithium iodide (9.56 g, 71.40 mmol). The reaction was heated at reflux for 4 h and then cooled before water (25 mL) was added and the mixture was acidified with 1 M HCl_(aq) to ~pH 3. The solution was extracted with EtOAc $(3 \times 25 \text{ mL})$, washed with water (25 mL), and dried over $MgSO₄$ before the volatiles were removed in vacuo to yield 3,3,3-trifluoro-2-(phenylamino)propanoic acid 4a as a brown solid (2.72 g, 85%): mp 101−103 °C; ¹ H NMR (400 MHz, CDCl3) δ 7.29−7.25 (m, 2H), 6.92−6.90 (m, 1H), 6.77−6.75 (m, 2H), 4.70 (q, J = 7.0 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 170.6, 144.7, 129.6, 123.1 (q, J = 283.5 Hz), 120.5, 114.3, 59.6 (q, J = 32.0 Hz); ¹⁹F NMR (376.5 MHz, CDCl₃) δ -72.4 (d, J = 7.0 Hz); FTIR 2994 (w), 1624 (m), 1520 (m), 1260 (s), 1181 (s), 1145 (s), 1031 (m), 778 (m), 658 (m) cm[−]¹ HRMS m/z [MH]⁺ calcd for $C_9H_9F_3NO_2$ 220.0585, found 220.0587.

Synthesis of 3,3,3-Trifluoro-2-(p-methoxyphenylamino) propanoic Acid 4b. To a stirring solution of methyl 3,3,3 trifluoro-2-(p-methoxyphenylamino)propanoate 3b (8.02 g, 30.47 mmol) in EtOAc (100 mL), shielded from light, was added lithium iodide (16.31 g, 121.88 mmol). The reaction was heated at reflux for 4 h and then cooled before water (25 mL) was added and the mixture was acidified with 1 M HCl_(aq) to ~pH 3. The solution was extracted with EtOAc $(3 \times 25 \text{ mL})$, washed with water (25 mL) , and dried over MgSO4 before the volatiles were removed in vacuo to yield 3,3,3 trifluoro-2-(p-methoxyphenylamino)propanoic acid 4b as a brown solid (5.76 g, 76%): mp 120−123 °C; ¹H NMR (400 MHz, CDCl₃) δ 6.84 (d, $J = 9.0$ Hz, 2H), 6.74 (d, $J = 9.0$ Hz, 2H), 6.52 (br, 1H), 4.56 $(q, J = 7.0 \text{ Hz}, 1H), 3.78 \text{ (s, 3H)}$; ¹³C NMR (100.6 MHz, CDCl₃) δ 170.2, 154.2, 138.6, 122.5 (q, J = 283.5 Hz), 116.4, 115.1, 61.0 (q, J = 31.5 Hz), 55.7; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -72.5 (d, J = 7.0) Hz); HRMS m/z [MH]⁺ calcd for C₁₀H₁₀F₃NO₃ 249.1865, found 249.1869.

Synthesis of 4-Trifluoromethyl-N-phenylsydnone 5a. To a stirring solution of 3,3,3-trifluoro-2-(phenylamino)propanoic acid 4a (3.20 g, 14.60 mmol) in dimethyoxyethane (15 mL) was added isoamyl nitrite (2.16 mL, 16.06 mmol). The reaction was stirred for 5 h and then concentrated in vacuo to yield an oil that was triturated with petroleum ether, filtered, and dried in vacuo to be used without further purification. (CAUTION: this nitrosamine intermediate is a suspected carcinogen). To a stirring suspension of the solid in DCM (15 mL) was added trifluoroacetic anhydride (4.06 mL, 29.20 mmol) at 0 °C. The reaction was warmed to ambient temperature and stirred for 1 h, and then approximately 10 mL of water was added and the mixture was neutralized with NaHCO₃. The organic layer was extracted with DCM (3×25 mL) and dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by recrystallization from ethanol to yield 4-trifluoromethyl-N-phenylsydnone 5a as a tan solid (2.31 g, 69%).

Note: Compound 5a could alternatively be purified by flash chromatography on silica gel (eluting with 20% EtOAc in petroleum ether): mp 112−114 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.78 (t, J = 7.5 Hz, 1H), 7.69 (t, J = 7.5 Hz, 2H), 7.62 (d, J = 7.5 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 163.2, 133.5, 133.3, 130.2, 124.6, 119.3 $(q, J = 268.0 \text{ Hz})$, 98.1 $(q, J = 41.0 \text{ Hz})$; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -56.3; FTIR 1791 (s), 1479 (m), 1330 (m), 1215 (m), 1120 (s), 1024 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₉H₆F₃N₂O₂ 231.0381, found 231.0389.

Synthesis of 4-Trifluoromethyl-N-(p-methoxyphenyl) **sydnone 5b.** To a stirring solution of $3,3,3$ -trifluoro-2- $(p$ methoxyphenylamino)propanoic acid 4b (4.75 g, 19.08 mmol) in dimethyoxyethane (20 mL) was added isoamyl nitrite (2.81 mL, 20.99 mmol). The reaction was stirred for 3 h and then concentrated in vacuo to yield an oil that was triturated with petroleum ether, filtered, and dried in vacuo to be used without further purification. (CAUTION: this nitrosamine intermediate is a suspected carcinogen.) To a stirring suspension of the solid in DCM (20 mL) was added trifluoroacetic anhydride (5.30 mL, 38.16 mmol) at 0 °C. The reaction was warmed to ambient temperature and stirred for 1 h, then approximately 10 mL of water was added and the mixture was neutralized with NaHCO₃. The organic layer was extracted with DCM $(3 \times 25 \text{ mL})$ and dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by recrystallization from ethanol to yield 4-trifluoromethyl- $N-(p$ -methoxyphenyl)sydnone 5b as a tan solid (2.78 g, 56%).

Note: Compound 5b could alternatively be purified by flash chromatography on silica gel (eluting with 20% EtOAc in petroleum ether): mp 65−66 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.54 (d, J = 9.0 Hz, 2H), 7.12 (d, J = 9.0 Hz, 2H), 3.94 (s, 3H); 13C NMR (100.6 MHz, CDCl₃) δ 163.2, 163.1, 125.9, 119.5 (q, J = 264.0 Hz), 115.5, 115.2, 98.0 (q, J = 43.0 Hz), 55.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −56.4; FTIR 2948 (w), 1790 (s), 1512 (m), 1476 (m), 1205 (m), 1124 (s), 1018 (s), 979 (m), 837 (m) cm[−]¹ ; HRMS m/ z [MH]⁺ calcd for $C_{10}H_8F_3N_2O_3$ 261.0487, found 261.0481.

General Procedure 1: Trifluoromethylation of Iodosydnones (See Scheme 4). To a stirring suspension of the iodosydnone (1 equiv) and CuI (1 equiv) in DMF (0.2 M) was added methyl fluorosulfonyldifluoroacetate (5 equiv) under a nitrogen atmosphere, and the mixtur[e](#page-1-0) was heated at 80 °C for the designated time. The reaction was filtered, brine was added to the filtrate, and the solution was extracted with EtOAc. The organic fractions were subsequently washed with brine and dried over $MgSO₄$ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 20% EtOAc in petroleum ether, unless otherwise stated) to yield the title sydnones.

4-Trifluoromethyl-N-phenylsydnone 5a. Following general procedure 1 using 4-iodo-N-phenylsydnone 7a (1.44 g, 5.00 mmol), CuI (920 mg, 5.00 mmol), and methyl fluorosulfonyldifluoroacetate (4.80 g, 25.00 mmol) in DMF (25 mL), stirring for 20 h, 4 trifluoromethyl-N-phenylsydnone 5a was isolated as a tan solid (906 mg, 79%).

4-Trifluoromethyl-N-(p-methoxyphenyl)sydnone 5b. Following general procedure 1 using 4-iodo-N-(para-methoxyphenyl) sydnone 7b (1.59 g, 5.0 mmol), CuI (952 mg, 5.0 mmol), and methyl fluorosulfonyldifluoroacetate (4.80 g, 25.00 mmol) in DMF (25 mL), stirring for 20 h, 4-trifluoromethyl-N-(p-methoxyphenyl) sydnone 5b was isolated as a tan solid (1.04 g, 80%).

4-Trifluoromethyl-N-(p-nitrophenyl)sydnone 5c. Following general procedure 1 using 4-iodo-N-(p-nitrophenyl)sydnone 7c (1.33 g, 4.00 mmol), CuI (762 mg, 4.00 mmol), and methyl fluorosulfonyldifluoroacetate (3.84 g, 20 mmol) in DMF (20 mL), stirring for 40 h, 4-trifluoromethyl- $N-(p$ -nitrophenyl)sydnone 5c was isolated as a yellow solid (605 mg, 55%): mp 110−111 °C; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 8.58 (d, J = 9.0 Hz, 1H), 7.92 (d, J = 9.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 162.5, 150.5, 137.7, 126.4, 125.7, 120.5 (q, J = 268.0 Hz), 98.4 (q, J = 43.0 Hz); ¹⁹F NMR (376.5 MHz, CDCl₃) δ –56.0; FTIR 3125 (w), 1780 (s), 1531 (s), 1349 (s), 1210 (s), 1119 (s), 1028 (s), 851 (s), 764 (m) cm⁻¹; HRMS m/ z $[EI]^+$ calcd for $C_9H_4F_3N_3O_4$ 275.0154, found 275.0165.

4-Trifluoromethyl-N-methylsydnone 5d. Following general procedure 1 using 4-iodo-N-methylsydnone 7d (1.13 g, 5.00 mmol), CuI (952 mg, 5.00 mmol), and methyl fluorosulfonyldifluoroacetate (4.80 g, 25.00 mmol) in DMF (25 mL), stirring for 20 h, 4 trifluoromethyl-N-methylsydnone 5d was isolated as a colorless oil (605 mg, 72%): ¹H NMR (400 MHz, CDCl₃) δ 4.23 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 163.2, 119.7 (q, J = 267.5 Hz), 96.9 (q, J $= 44.0$ Hz), 40.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –58.3; FTIR 3051 (w), 1774 (s), 1516 (m), 1353 (m), 1196 (s), 1126 (s), 1004 (s), 743 (m), 600 (m) cm⁻¹; HRMS m/z [EI] ⁺ calcd for $C_4H_4F_3N_2O_2$ 169.0225, found 169.0228.

4-Trifluoromethyl-N-benzylsydnone 5e. Following general procedure 1 using 4-iodo-N-benzylsydnone 7e (1.51 g, 5.00 mmol), CuI (952 mg, 5.00 mmol), and methyl fluorosulfonyldifluoroacetate (4.80 g, 25.00 mmol) in DMF (25 mL), stirring for 20 h, 4 trifluoromethyl-N-benzylsydnone 5e was isolated as a yellow solid (580 mg, 48%): mp 50−51 °C; ¹ H NMR (400 MHz, CDCl3) δ 7.46− 7.44 (m, 3H), 7.39-7.38 (m, 2H), 5.61 (s, 2H, CH2); ¹³C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 163.4, 130.3, 129.4, 129.3, 128.5, 119.6 $(q, J =$ 267.5 Hz), 96.2 (q, J = 43.5 Hz), 58.1; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.2; FTIR 2919 (w), 1768 (s), 1513 (m), 1383 (m), 1173 (m), 1114 (s), 1021 (s), 729 (m), 695 (s) cm⁻¹; HRMS m/z [EI]⁺ calcd for $C_{10}H_7F_3N_2O_2$ 244.0460, found 244.0459.

General Procedure 2: Cycloaddition Reactions of 4- Trifluoromethylsydnones and Alkynes (See Table 1). A solution of a 4-trifluoromethylsydnone (1 equiv) and an alkyne (2 equiv) in o-DCB (0.5 M) in a sealed microwave vessel was heated at 180 °C for the designated length of time. After cooling, the crude [ma](#page-2-0)terials were purified by flash chromatography on silica gel (eluting with 5% EtOAc in petroleum ether, unless otherwise stated) to yield the title pyrazoles.

Synthesis of 1,3-Diphenyl-5-trifluoromethyl-1H-pyrazole 8a. Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmol), heating for 24 h, 1,3-diphenyl-5-trifluoromethyl-1H-pyrazole 8a was isolated as a colorless solid (125 mg, 87%): mp 50−53 °C (lit.⁹ mp 50−52 °C); ¹H NMR (400 MHz, CDCl₃) *δ* 7.92−7.90 (m, 2H), 7.62−7.39 (m, 8H), 7.16 (s, 1H); 13C NMR (100.6 MHz, CDCl3) [δ](#page-14-0) 151.7, 139.2, 134.0 (q, J = 39.0 Hz), 131.8, 129.3, 129.2, 128.8, 128.7, 125.9, 125.8, 119.8 (q, J = 269.0 Hz), 106.1 (br); ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.6; FTIR 3073 (w), 1600 (w), 1505 (m), 1447 (m), 1288 (m), 1237 (s), 1167 (s), 989 (m), 770 (s), 693 (s) cm⁻¹; HRMS m / z [MH]⁺ calcd for C₁₆H₁₂F₃N₂ 289.0953, found 289.0943.

Synthesis of 1-Phenyl-3-n-butyl-5-trifluoromethyl-1H-pyrazole 9a.10 Following general procedure 2 using 4-trifluoromethyl-Nphenylsydnone 5a (115 mg, 0.5 mmol) and 1-hexyne (82 mg, 1.0 mmol), [he](#page-14-0)ating for 24 h, 1-phenyl-3-n-butyl-5-trifluoromethyl-1Hpyrazole **9a** was isolated as a yellow oil (105 mg, 78%): ¹H NMR (400 MHz, CDCl₃) δ 7.49–7.44 (m, 5H), 6.64 (s, 1H), 2.73 (t, J = 7.5 Hz, 2H), 1.76−1.68 (m, 2H), 1.50−1.41 (m, 2H), 0.98 (m, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 153.8, 139.3, 132.9 (q, J = 39.0 Hz), 129.0, 128.9, 125.6, 119.7 (q, J = 269.0 Hz), 107.7 (br), 31.5, 27.7, 22.4, 13.8; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -57.5; FTIR 3060 (w), 2968 (m), 2939 (m), 2862 (w), 1600 (m), 1505 (s), 1470 (m), 1390 (w), 1295 (m), 1186 (s), 1129 (s), 989 (m), 763 (m), 693 (m) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for C₁₄H₁₆F₃N₂ 269.1269, found 269.1266.

Synthesis of 1-Phenyl-3-trimethylsilyl-5-trifluoromethyl-1Hpyrazole 10a. Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and ethynyltrimethylsilane (98 mg, 1.0 mmol), heating for 24 h, 1-phenyl-3-trimethylsilyl-5 trifluoromethyl-1H-pyrazole 10a was isolated as a colorless oil (106 mg, 75%): ¹H NMR (400 MHz, CDCl₃) δ 7.52−7.48 (m, 5H), 6.92 (s, 1H), 0.38 (s, 9H); ¹³C NMR (100.6 MHz, CDCl₃) δ 153.9, 139.4, 132.9 (q, $J = 39.0$ Hz), 129.1, 129.0, 125.8, 120.4 (q, $J = 269.0$ Hz), 114.8 (br), -1.2 ; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.1; FTIR 3088 (w), 2960 (m), 1604 (m), 1504 (m), 1283 (m), 1171 (s), 975 (m), 850 (s), 761 (m), 689 (m) cm[−]¹ ; HRMS m/ z [MH]⁺ calcd for $C_{13}H_{16}F_3N_2Si$ 285.1035, found 285.1047.

Synthesis of 2-(1-Phenyl-5-trifluoromethyl-1H-pyrazol-3-yl) pyridine 11a.¹¹ Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 2-ethynylpyridine (103 mg, 1.0 mmol), heating for 24 h, 2-(1-phenyl-5 trifluoromethyl-1H-pyrazol-3-yl)pyridine 11a was isolated as a yellow oil (121 mg, 84%): ¹H NMR (400 MHz, CDCl₃) δ 8.68 (ddd, J = 1.0, 5.0, 7.5 Hz, 1H), 8.07 (td, $J = 1.0$, 8.0 Hz, 1H), 7.76 (td, $J = 2.0$, 7.5 Hz, 1H), 7.59–7.51 (m, 6H), 7.28 (ddd, J = 1.0, 5.0, 7.5 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 152.0, 150.8, 149.6, 139.2, 136.7, 134.2 $(q, J = 39.5 \text{ Hz})$, 129.4, 129.1, 125.8, 123.3, 120.2, 119.7 $(q, J = 269.0 \text{ s})$ Hz), 107.7 (br); ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.7; FTIR 3076 (w), 1599 (w), 1499 (m), 1415 (m), 1282 (m), 1232 (s), 1136 (s), 994 (m), 732 (s), 694 (s) cm⁻¹; HRMS *m/ z* [MH]⁺ calcd for $C_{15}H_{11}F_3N_3$ 290.0905, found 290.0909.

Synthesis of 1-Phenyl-3-(cyclohex-1-enyl)-5-trifluoromethyl-1H-pyrazole 12a. Following general procedure 2 using 4 trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 2 ethynylcyclohexene (106 mg, 1.0 mmol), heating for 24 h, 1-phenyl-3- (cyclohex-1-enyl)-5-trifluoromethyl-1H-pyrazole 12a was isolated as a yellow solid (130 mg, 89%): mp 55−57 °C; ¹H NMR (400 MHz, CDCl3) δ 7.53−7.46 (m, 5H), 6.86 (s, 1H), 6.45−6.43 (m, 1H), 2.53−2.50 (m, 2H), 2.27−2.22 (m, 2H), 1.83−1.77 (m, 2H), 1.74− 1.68 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 153.6, 139.4, 133.1 $(q, J = 42.5 \text{ Hz})$, 129.6, 129.0 (2), 126.5, 127.0, 119.9 (q, $J = 269.0$ Hz), 105.0 (br), 25.8, 25.5, 22.5, 22.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.6; FTIR 3063 (w), 2939 (m), 2860 (w), 1600 (m), 1505 (s), 1450 (m), 1288 (m), 1231 (s), 1139 (s), 989 (m), 817 (w), 771 (m), 697 (m) cm⁻¹; HRMS *m/ z* [MH]⁺ calcd for $C_{16}H_{16}F_3N_2$ 293.1266, found 293.1267.

Synthesis of 1-Phenyl-3-cyclopropyl-5-trifluoromethyl-1Hpyrazole 13a. Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and ethynylcyclopropane (66 mg, 1.0 mmol), heating for 24 h, 1-phenyl-3-cyclopropyl −5 trifluoromethyl-1H-pyrazole 13a was isolated as a yellow oil (111 mg, 88%): ¹ H NMR (400 MHz, CDCl3) δ 7.50−7.46 (m, 5H), 6.50 (s, 1H), 2.03 (tt, J = 5.0, 8.5 Hz, 1H), 1.03−0.99 (m, 2H), 0.87−0.82 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 155.5, 139.3, 132.9 (q, J = 39.0 Hz), 129.0, 128.9, 125.6, 119.8 (q, J = 269.0 Hz), 105.6 (br), 8.9, 8.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.6; FTIR 3060 (w), 2936 (m), 2856 (w), 1603 (w), 1508 (m), 1288 (m), 1234 (m), 1177 (m), 1136 (s), 989 (m), 771 (m), 690 (m) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for $C_{13}H_{12}F_{3}N_{2}$ 253.0953, found 253.0957.

Synthesis of 3-(Benzyloxymethyl)-1-phenyl-5-trifluoromethyl-1H-pyrazole 14a. Following general procedure 2 using 4 trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and benzyl 2-propynyl ether (146 mg, 1.0 mmol), heating for 24 h, 3- (benzyloxymethyl)-1-phenyl-5-trifluoromethyl-1H-pyrazole 14a was isolated as a yellow oil (139 mg, 84%): ^1H NMR (400 MHz, CDCl₃) δ 7.54–7.39 (m, 10H), 6.93 (s, 1H), 4.69 (s, 2H), 4.69 (s, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 150.3, 139.1, 137.9, 133.6 (q, J $= 39.0$ Hz), 129.3, 129.1, 128.5, 128.0, 127.8, 125.7, 119.8 (q, J = 269.0 Hz), 108.4 (br), 72.8, 65.5; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.6; FTIR 3069 (w), 3034 (w), 2870 (m), 1732 (w), 1507 (s), 1299 (s), 1188 (s), 1084 (s), 992 (m), 736 (m), 701 (s) cm⁻¹; HRMS m/z [MH] ⁺ calcd for $C_{18}H_{16}F_3N_2O_2$ 333.1215, found 333.1228.

Synthesis of 1-Phenyl-3-(3-chloro)propyl-5-trifluoromethyl-1H-pyrazole 15a. Following general procedure 2 using 4 trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 5 chloro-1-pentyne (103 mg, 1.0 mmol), heating for 24 h, 1-phenyl-3-(3 chloro)propyl-5-trifluoromethyl-1H-pyrazole 15a was isolated as a yellow oil (101 mg, 70%): ¹H NMR (400 MHz, CDCl₃) δ 7.49 (m, 5H), 6.67 (s, 1H), 3.65 (t, J = 6.5 Hz, 2H), 2.91 (t, J = 7.0 Hz, 2H), 2.25−2.18 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 151.8, 139.2, 133.1 (q, $J = 39.5$ Hz), 129.1, 129.0, 125.6, 119.8 (q, $J = 269.0$ Hz), 107.9 (br), 44.2, 31.9, 25.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −57.5; FTIR 3063 (w), 2958 (w), 2869 (w), 1600 (m), 1508 (s), 1470 (m), 1295 (m), 1183 (s), 1136 (s), 989 (m), 817 (w), 767 (m), 690 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₃H₁₃ClF₃N₂ 289.0719, found 289.0708.

Synthesis of 1,4-Diphenyl-4-n-butyl-5-trifluoromethyl-1Hpyrazole 16a and 1,3-Diphenyl-3-n-butyl-5-trifluoromethyl-1H-pyrazole 16b. Following general procedure 2 using 4trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 1phenyl-1-hexyne (158 mg, 1.0 mmol), heating for 24 h, 1,4-diphenyl-4 n-butyl-5-trifluoromethyl-1H-pyrazole 16a and 1,3-diphenyl-3-n-butyl-5-trifluoromethyl-1H-pyrazole 16b were isolated as an inseparable mixture (48:52, 16a/16b) as a yellow solid (103 mg, 62%): $^1\rm H$ NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 7.68–7.66 (m, 1H), 7.58–7.42 (m, 8H), 7.39– 7.37 (m, 1H), 2.78 (t, J = 7.5 Hz, 0.96H), 2.64 (t, J = 7.5 Hz, 1.04H), 1.67−1.57 (m, 2H), 1.46−1.29 (m, 2H), 0.95 (t, J = 7.5 Hz, 1.44H), 0.87 (t, J = 7.5 Hz, 1.56H); ¹³C NMR (100.6 MHz, CDCl₃) δ 152.5, 151.8, 140.0, 139.9, 132.6, 131.2, 130.0, 129.1, 129.0, 128.9 (2), 128.6, 128.5, 128.2, 127.8, 126.2, 126.0, 123.9, 122.7, 120.8 (q, J = 269.0 Hz), 120.4 (q, J = 269.0 Hz), 33.6, 31.3, 26.0, 23.2, 22.7, 22.5, 13.8, 13.7; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −54.3 (0.52F), −55.5 (0.48F); FTIR 3061 (w), 2961 (w), 2877 (w), 1594 (w), 1503 (m), 1319 (m), 1174 (s), 1117 (s), 979 (m), 773 (m), 692 (s), 524 (m) cm⁻¹; HRMS m/z [MH] ⁺ calcd for C₂₀H₂₀F₃N₂: 345.1579, found 345.1566 and 345.1570.

Synthesis of Methyl 3-Methyl-1-phenyl-5-trifluoromethyl-1H-pyrazole-4-carboxylate 17a and Methyl 4-Methyl-1-phenyl-5-trifluoromethyl-1H-pyrazole-3-carboxylate 17b. Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and methyl-2-butynoate (196 mg, 2.0 mmol), heating for 24 h, eluting with (10% EtOAc in petroleum ether), 3 methyl-1-phenyl-5-trifluoromethyl-1H-pyrazole-4-carboxylate 17a was isolated as a yellow solid (217 mg, 76%) and methyl 4-methyl-1 phenyl-5-trifluoromethyl-1H-pyrazole-3-carboxylate 17b was isolated as a yellow solid (39 mg, 14%).

17a: mp 78-79 °C (lit.¹² mp 81-83 °C); ¹H NMR (400 MHz, CDCl3) δ 7.50−7.47 (m, 3H), 7.42−7.39 (m, 2H), 3.91 (s, 3H), 2.50 (s, 3H); 13C NMR (100.6 [MH](#page-14-0)z, CDCl3) δ 162.6, 151.6, 139.4, 133.1 $(q, J = 39.5 \text{ Hz})$, 129.7, 129.1, 125.8, 119.1 $(q, J = 271.5 \text{ Hz})$, 114.1, 52.1, 13.5; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –55.8; FTIR 3048 (w), 2950 (w), 1720 (s), 1553 (m), 1317 (m), 1250 (s), 1170 (s), 1140 (s), 1098 (s), 999 (m), 959 (w), 785 (s) cm[−]¹ ; HRMS m/ z [MH]⁺ calcd for $C_{13}H_{12}F_3N_2O_2$ 285.0851, found 285.0849.

17b: mp 65–67 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.50–7.47 (m, 3H), 7.44−7.42 (m, 2H), 3.95 (s, 3H), 2.53 (q, J = 2.0 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 162.4, 141.9, 139.2, 131.1 (q, J = 37.5) Hz), 129.9, 129.1, 126.3, 124.2, 120.1, $(q, J = 270.5 \text{ Hz})$, 52.1, 8.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –55.9; FTIR 3059 (w), 2959 (w), 1719 (s), 1501 (m), 1230 (s), 1187 (s), 1118 (s), 1043 (s), 980 (m), 776 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₃H₁₂F₃N₂O₂ 285.0851, found 285.0846.

Synthesis of 1,3,4-Triphenyl-5-trifluoromethyl-1H-pyrazole 18. Following general procedure 2 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 1,2-diphenylethyne (178 mg, 1.0 mmol), heating for 24 h, 1,3,4-triphenyl-5-trifluoromethyl-1H-pyrazole ¹⁸ was isolated as an off-white solid (97 mg, 53%): mp 139−¹⁴⁰ °C. ¹ ¹H NMR (400 MHz, CDCl₃) δ 7.65–7.63 (m, 2H), 7.58–7.53 (m, 3H), 7.46−7.42 (m, 5H), 7.38−7.35 (m, 2H), 7.28−7.26 (m, 3H); 13C NMR (100.6 MHz, CDCl₃) δ 150.4, 139.8, 131.7, 131.1, 130.4 (q, J = 36.5 Hz), 129.3, 129.1, 128.4, 128.3, 128.2, 128.0, 126.1, 122.3, 120.1 (q, $J = 271.0$ Hz), unable to unequivocally assign 2 \times CH signals; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –54.3; FTIR 1504 (m), 1443 (m), 1224 (m), 1171 (s), 1121 (s), 976 (m), 777 (m), 689 (s), 547 (m) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for $C_{22}H_{16}F_3N_2$ 365.1266, found 365.1257.

Synthesis of 1-(p-Methoxyphenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole 19a. Following general procedure 2 using 4 trifluoromethyl-N-(p-methoxyphenyl)sydnone 5b (130 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmol), heating for 24 h, $1-(p$ methoxyphenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole 19a was isolated as a yellow solid (135 mg, 85%): mp 89−90 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.91–7.87 (d, J = 9.0 Hz, 2H), 7.51–7.39 (m, 5H), 7.11 (s, 1H), 7.06−7.01 (d, J = 9.0 Hz, 2H), 3.89 (s, 3H); 13C NMR (100.6 MHz, CDCl₃) δ 160.2, 151.4, 134.0 (q, J = 39.0 Hz), 132.2, 131.9, 128.8, 128.6, 127.3, 125.9, 119.9 (q, J = 269.0 Hz), 114.3, 105.6, 55.6; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.9; FTIR 3066 (w), 1512 (m), 1236 (m), 1166 (s), 1124 (s), 1028 (m), 986 (m), 838 (m) cm⁻¹; HRMS *m/z* [MH] ⁺ calcd for C₁₇H₁₄F₃N₂O 319.1058, found 319.1053.

Synthesis of 1-(p-Methoxyphenyl)-3-n-butyl-5-trifluoromethyl-1H-pyrazole 20a. Following general procedure 2 using 4 trifluoromethyl-N-(p-methoxyphenyl)sydnone 5b (130 mg, 0.5 mmol) and 1-hexyne $(82 \text{ mg}, 1.0 \text{ mmol})$, heating for 24 h, 1- $(p$ methoxyphenyl)-3-n-butyl-5-trifluoromethyl-1H-pyrazole 20a was isolated as a yellow oil (106 mg, 71%): ^1H NMR (400 MHz, CDCl₃) δ 7.39 (d, J = 9.0 Hz, 2H), 6.97 (d, J = 9.0 Hz, 2H), 6.59 (s, 1H), 3.86 (s, 3H), 2.71 (t, J = 7.5 Hz, 2H), 1.74−1.66 (m, 2H), 1.48−1.39 (m, 2H), 0.97 (t, J = 7.5 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 160.0, 153.4, 133.0 (q, $J = 38.5$ Hz), 132.3, 127.1, 180.6 (q, $J = 269.0$ Hz), 114.1, 107.1, 55.5, 31.6, 27.7, 22.4, 13.8; 19F NMR (376.5 MHz, CDCl₃) δ –57.8; FTIR 2956 (m), 2931 (m), 2860 (w), 1523 (s), 1466 (m), 1250 (s), 1186 (s), 1125 (s), 990 (m), 837 (m) cm⁻¹; HRMS m / z [MH]⁺ calcd for C₁₅H₁₈F₃N₂O 299.1371, found 299.1375.

Synthesis of 1-(p-Methoxyphenyl)-3-cyclopropyl-5-trifluoromethyl-1H-pyrazole 21a. Following general procedure 2 using 4 trifluoromethyl-N-(p-methoxyphenyl)sydnone 5b (130 mg, 0.5 mmol) and ethynylcyclopropane (102 mg, 1.0 mmol), heating for 24 h, 1- (para-methoxyphenyl)-3-cyclopropyl-5-trifluoromethyl-1H-pyrazole $21a$ was isolated as a yellow oil $(106 \text{ mg}, 75\%)$: ¹H NMR (400 MHz, CDCl₃) δ 7.38 (d, J = 9.0 Hz, 2H), 6.97 (d, J = 9.0 Hz, 2H), 6.45 (s, 1H), 3.87 (s, 3H), 2.01 (tt, J = 5.0, 8.5 Hz, 1H), 1.02−0.97 (m, 2H), 0.84−0.80 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 160.0, 155.1, 133.1 (q, $J = 38.5$ Hz), 132.2, 127.2, 119.8 (q, $J = 269.0$ Hz), 114.1, 105.0, 55.5, 8.91, 8.16; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.9; FTIR 3090 (w), 3010 (w), 2840 (w), 1519 (s), 1485 (m), 1253 (s), 1177 (s), 1132 (s), 1033 (m), 986 (m), 836 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{14}H_{14}F_3N_2O$ 283.1058, found 283.1057.

Synthesis of 1-(p-Nitrophenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole 22a. Following general procedure 2 using 4 trifluoromethyl-N- $(p$ -nitrophenyl)sydnone 5c (138 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmol), heating for 8 h, 1-(pnitrophenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole 22a was isolated as a yellow solid (141 mg, 85%): mp 97–99 °C (lit.⁸ mp 102 °C); ¹H NMR (400 MHz, CDCl₃) δ 8.42 (d, J = 9.0 Hz, 2H), 7.90–7.88 (m, 2H), [7](#page-14-0).84 (d, J = 9.0 Hz, 2H), 7.51–7.42 (m, 3H), 7.23 (s, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 152.9, 147.5, 144.0, 134.0 (q, J = 39.5 Hz), 131.0, 129.3, 129.0, 126.0, 125.6, 124.7, 119.6 (q, J = 269.5 Hz), 108.0 (br); ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.1; FTIR 3100 (w), 1596 (m), 1517 (m), 1346 (s), 1229 (s), 1124 (s), 987 (m), 853 (s), 774 (s), 690 (s) cm⁻¹; HRMS m/z [MH] ⁺ calcd for $C_{16}H_{11}F_3N_3O_2$ 334.0803, found 334.0802.

Synthesis of 1-(p-Nitrophenyl)-3-(3-chloro)propyl-5-trifluoromethyl-1H-pyrazole 23a. Following general procedure 2 using 4 trifluoromethyl-N-(p-nitrophenyl)sydnone 5c (138 mg, 0.5 mmol) and 5-chloro-1-pentyne (103 mg, 1.0 mmol), heating for 8 h, $1-(p$ nitrophenyl)-3-(3-chloro)propyl-5-trifluoromethyl-1H-pyrazole 23a was isolated as a yellow solid (114 mg, 68%): mp 50–51 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.38 (d, J = 9.0 Hz, 2H), 7.74 (d, J = 9.0 Hz, 2H), 6.77 (s, 1H), 3.66 (t, $J = 6.5$ Hz, 2H), 2.93 (t, $J = 7.5$ Hz, 2H), 2.26−2.19 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 153.4, 147.4, 143.9, 133.3 (q, $J = 39.5$ Hz), 125.4, 124.7, 119.6 (q, $J = 269.0$ Hz), 110.0 (br), 44.1, 31.6, 26.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −57.1; FTIR 3140 (W), 2963 (w), 2850 (w), 1601 (m), 1527 (s), 1350 (s), 1220 (m), 1125 (s), 991 (m), 857 (s), 692 (m) cm⁻¹; HRMS m / z [MH]⁺ calcd for C₁₃H₁₂ClF₃N₃O₂ 334.0582, found 334.0570.

Synthesis of 1-Methyl-3-phenyl-5-trifluoromethyl-1H-pyra**zole 24a.**⁸ Following general procedure 2 using 4-trifluoromethyl-Nmethylsydnone 5d (84 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmo[l\),](#page-14-0) 1-methyl-3-phenyl-5-trifluoromethyl-1H-pyrazole 24a was isolated as a yellow oil (107 mg, 95%): ¹H NMR (400 MHz, CDCl₃) δ 7.83−7.81 (m, 2H), 7.47−7.43 (m, 2H), 7.40−7.35 (m, 1H), 6.93 (s, 1H), 4.05 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 150.4, 133.1 (q, J $=$ 39.0 Hz), 132.2, 128.8, 128.3, 125.6, 120.1 (q, J = 268.5 Hz), 104.5, 38.1; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –60.5; HRMS m/z [MH]⁺ calcd for $C_{11}H_{10}F_3N_2$ 227.0796, found 227.0807.

Synthesis of 1-Methyl-3-(cyclohex-1-enyl)-5-trifluoromethyl-1H-pyrazole 25a. Following general procedure 2 using 4trifluoromethyl-N-methylsydnone 5d (84 mg, 0.5 mmol) and 2 ethynylcyclohexene (106 mg, 1.0 mmol), heating for 24 h, 1-methyl-3- (cyclohex-1-enyl)-5-trifluoromethyl-1H-pyrazole 25a was isolated as a yellow oil (94 mg, 82%): ¹H NMR (400 MHz, CDCl₃) δ 6.63 (s, 1H), 6.33−6.30 (m, 1H), 3.96 (s, 3H), 2.45−2.41 (m, 2H), 2.23−2.18 (m, 2H), 1.80−1.73 (m, 2H), 1.71−1.66 (m, 2H); 13C NMR (100.6 MHz, CDCl₃) δ 152.4, 132.4 (q, J = 39.0 Hz), 129.5, 125.5, 120.2 (q, J = 268.5 Hz), 103.3 (br), 37.8, 25.8, 25.4, 22.5, 22.2; 19F NMR (376.5 MHz, CDCl₃) δ –60.5; FTIR 3138 (w), 2925 (m), 2864 (w), 1458 (m), 1270 (s), 1188 (m), 1127 (s), 1049 (s), 813 (m), 718 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₁H₁₄F₃N₂ 231.1109, found 231.1100.

Synthesis of 3-(Benzyloxymethyl)-1-methyl-5-trifluoromethyl-1H-pyrazole 26a. Following general procedure 2 using 4 trifluoromethyl-N-methylsydnone 5d (84 mg, 0.5 mmol) and benzyl 2 propynyl ether (146 mg, 1.0 mmol), heating for 24 h, 3- (benzyloxymethyl)-1-methyl-5-trifluoromethyl-1H-pyrazole 26a was isolated as a yellow oil (120 mg, 89%): $^1\mathrm{H}$ NMR (400 MHz, CDCl₃) δ 7.39–7.33 (m, 4H), 7.34–7.31 (m, 1H), 6.68 (s, 1H), 4.61 (s, 2H), 4.56 (s, 2H), 3.99 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 148.7, 137.8, 132.8 (q, J = 39.0 Hz), 128.5, 128.0, 127.8, 120.0 (q, J = 269.0 Hz), 107.0 (br), 72.6, 65.3, 38.0; 19F NMR (376.5 MHz, CDCl₃) δ -60.5; FTIR 3029 (w), 2950 (w), 2863 (w), 1458 (m), 1270 (s), 1178 (s), 1117 (s), 1038 (s), 824 (m), 736 (m), 697 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₃H ₁₄F₃N₂O 271.1058, found 271.1053.

Synthesis of Ethyl 1-Methyl-5-trifluoromethyl-1H-pyrazole-3-carboxylate 27a and Ethyl 1-Methyl-5-trifluoromethyl-1Hpyrazole-4-carboxylate 27b.⁸ Following general procedure 2 using 4-trifluoromethyl-N-methylsydnone 5d (84 mg, 0.5 mmol) and ethylpropiolate (98 mg, 1.0 [m](#page-14-0)mol), eluting with 15% EtOAc in petroleum ether, ethyl 1-methyl-5-trifluoromethyl-1H-pyrazole-3 carboxylate 27a and ethyl 1-methyl-5-trifluoromethyl-1H-pyrazole-4 carboxylate 27b were isolated as an inseparable mixture (93:7, 27a/ 27b) as a yellow oil (104 mg, 94%).

Spectroscopic data only reported for 27a: 1 H NMR (400) MHz, CDCl₃) δ 7.12 (s, 1H), 4.40 (q, J = 7.0 Hz, 2H), 4.06 (s, 3H), 1.38 (t, J = 7.0 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 161.1, 142.6, 133.3 (q, $J = 40.0$ Hz), 119.4 (q, $J = 269.0$ Hz), 110.3, 61.4, 38.9, 14.3; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –61.0; FTIR 3147 (w), 2986 (w), 1723 (m), 1447 (m), 1270 (s), 1214 (s), 1119 (s), 1046 (s), 994 (s), 825 (m), 783 (s), 709 (m) cm⁻¹; HRMS *m*/z [MH]⁺ calcd for $C_8H_{10}F_3N_2O_2$ 223.0694, found 223.0686.

Synthesis of 2-Trifluoromethyl-5,6-dihydro-4H-pyrrolo[1,2 **b]pyrazole 29.** Following general procedure 2 using 4-trifluoromethyl-N-methylsydnone 5d (84 mg, 0.5 mmol) and 5-chloro-1 pentyne (103 mg, 1.0 mmol), heating for 24 h, eluting with 30% EtOAc in petroleum ether, 2-trifluoromethyl-5,6-dihydro-4H-pyrrolo- [1,2-b]pyrazole 29 was isolated as a yellow oil (56 mg, 64%): 1 H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 6.25 (s, 1H), 4.20 (t, J = 7.0 Hz, 2H), 2.94 (t, J = 7.0 Hz, 2H), 2.70–2.62 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 147.0, 146.2 (q, $J = 37.5$ Hz), 121.2 (q, $J = 268.5$ Hz), 97.6 (br), 48.1, 26.1, 23.0; ¹⁹F NMR (376.5 MHz, CDCl₃) δ −62.1; FTIR 3143 (w), 2972 (w), 2880 (w), 1484 (m), 1239 (m), 1161 (s), 1117 (s), 1086 (s), 977 (s), 802 (m), 702 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_7H_8F_3N_2$, 177.0640, found 177.0645.

Synthesis of 1-Benzyl-3-phenyl-5-trifluoromethyl-1H-pyrazole 39a and 1-Benzyl-3- trifluoromethyl-5-phenyl-1H-pyrazole 39b. Following general procedure 2 using 4-trifluoromethyl-Nbenzylsydnone 5e (122 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmol), heating for 24 h, eluting with 5% ether in petroleum ether, 1-benzyl-3-phenyl-5-trifluoromethyl-1H-pyrazole 39a was isolated as yellow oil (59 mg, 39%) and 1-benzyl-3-trifluoromethyl-5-phenyl-1Hpyrazole 39b was isolated as a colorless solid (33 mg, 22%).

39a: ¹ H NMR (400 MHz, CDCl3) δ 7.85−7.83 (m, 2H), 7.47−7.43 (m, 2H), 7.40−7.32 (m, 4H), 7.30−7.28 (m, 2H), 6.97 (s, 1H), 5.50 $(s, 2H)$; ¹³C NMR (100.6 MHz, CDCl₃) δ 150.1, 135.9, 133.1 (q, J = 39.0 Hz), 132.1, 128.8, 128.7, 128.4, 128.1, 127.3, 125.7, 120.0 (q, J = 269.0 Hz), 104.9 (br), 54.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –59.2; FTIR 3067 (w), 3028 (w), 2930 (w). 1445 (m), 1269 (s), 1153 (s),

1128 (s), 1046 (s), 760 (m), 691 (s) cm[−]¹ ; HRMS m/z [MH] ⁺ calcd for $C_{17}H_{14}F_3N_2$ 303.1109, found 303.1108.

39b: mp 53–54 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.48–7.43 (m, 3H), 7.35−7.30 (m, 5H), 7.10−7.08 (m, 2H), 6.64 (s, 1H), 5.41 (s, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 145.6, 142.1 (q, J = 38.0 Hz), 136.4, 129.4, 129.3, 129.1, 128.9, 128.8, 127.9, 126.9, 121.5 (q, J = 268.5 Hz), 104.8 (br), 53.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –61.9; FTIR 3075 (w), 3031 (w), 1475 (m), 1211 (m), 1128 (s), 1106 (m), 989 (s), 758 (m), 692 (s) cm⁻¹; HRMS *m*/z [MH]⁺ calcd for $C_{17}H_{14}F_3N_2$ 303.1109, found 303.1096.

Synthesis of 1-Benzyl-3-n-butyl-5-trifluoromethyl-1H-pyrazole 40a and 1-Benzyl-3- trifluoromethyl-5-n-butyl-1H-pyrazole 40b. Following general procedure 2 using 4-trifluoromethyl-Nbenzylsydnone 5e (122 mg, 0.5 mmol) and 1-hexyne (82 mg, 1.0 mmol), heating for 24 h, 1-benzyl-3-n-butyl-5-trifluoromethyl-1Hpyrazole 40a was isolated as a yellow oil (49 mg, 35%) and 1-benzyl-3 trifluoromethyl-5-n-butyl-1H-pyrazole 40b was isolated as a yellow oil (19 mg, 13%).

40a: ¹ H NMR (400 MHz, CDCl3) δ 7.37−7.28 (m, 3H), 7.20−7.18 (m, 2H), 6.48 (s, 1H), 5.40 (s, 2H), 2.67 (t, J = 7.5 Hz, 2H), 1.70− 1.62 (m, 2H), 1.46−1.47 (m, 2H), 0.97 (t, J = 7.5 Hz, 3H); 13C NMR $(100.6 \text{ MHz}, \text{CDCl}_3) \delta 153.0, 136.3, 132.2 \text{ (q, } J = 39.0 \text{ Hz}), 128.6,$ 127.9, 127.1, 120.2 (q, J = 269.0 Hz), 106.4 (br), 54.4, 31.7, 27.7, 22.4, 13.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –59.1; FTIR 3033 (w), 2956 (w), 2931 (w), 2861 (w), 1565 (w), 1453 (m), 1264 (s), 1159 (s), 1127 (s), 1036 (s), 808 (m), 693 (m) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for $C_{15}H_{18}F_3N_2$ 283.1422, found 283.1415.

40b: ¹ H NMR (400 MHz, CDCl3) δ 7.37−7.31 (m, 3H), 7.12−7.10 $(m, 2H)$, 6.36 (s, 1H), 5.36 (s, 2H), 2.52 (t, J = 7.5 Hz, 2H), 1.59– 1.51 (m, 2H), 1.39−1.31 (m, 2H), 0.89 (t, J = 7.5 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 145.1, 141.5 (q, J = 37.5 Hz), 136.0, 128.9, 128.0, 126.8, 121.6 (q, J = 269.0 Hz), 103.2, 53.8, 30.2, 25.2, 22.2, 13.7; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –61.9; FTIR 3032 (w), 2964 (w), 2930 (w), 2847 (w), 1609 (m), 1494 (m), 1214 (s), 1120 (s), 984 (m), 806 (m), 725 (m) cm⁻¹; HRMS *m*/z [MH]⁺ calcd for $C_{15}H_{18}F_3N_2$ 283.1422, found 283.1415.

Synthesis of Dimethyl 1-Benzyl-5-trifluoromethyl-1H-pyrazole-3,4-dicarboxylate 45a and Dimethyl 1-Benzyl-3-trifluoromethyl-1H-pyrazole-4,5-dicarboxylate 45b. Following general procedure 2 using 4-trifluoromethyl-N-benzylsydnone 5e (122 mg, 0.5 mmol) and DMAD (142 mg, 1.0 mmol), heating at 120 °C for 24 h, dimethyl 1-benzyl-5-trifluoromethyl-1H-pyrazole-3,4-dicarboxylate 45a and dimethyl 1-benzyl-3-trifluoromethyl-1H-pyrazole-4,5-dicarboxylate 45b were isolated as an inseparable mixture (96:4, 45a/45b) as a yellow oil (93 mg, 54%).

Spectroscopic data only reported for 45a: 1 H NMR (400) MHz, CDCl₃) δ 7.37–7.34 (m, 3H), 7.26–7.24 (m, 2H), 5.54 (s, 2H), 3.96 (s, 3H), 3.94 (s, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 162.6, 160.5, 140.6, 134.0, 131.0 (q, J = 40.5 Hz), 128.9, 128.7, 127.5, 119.1, 118.8 (q, $J = 271.0$ Hz), 56.4, 53.2, 52.8; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –58.4; FTIR 3043 (w), 2956 (w), 1729 (s), 1496 (m), 1315 (m), 1250 (s), 1176 (s), 1142 (s), 1047 (s), 965 (m), 831 (m), 731 (m) cm⁻¹; HRMS m / z [MH]⁺ calcd for C₁₅H₁₄F₃N₂O₄ 343.0906, found 343.0922.

Synthesis of Dimethyl 1-Benzyl-5-trifluoromethyl-1H-pyrazole-3,4-dicarboxylate 45a and Dimethyl 1-Benzyl-3-trifluoromethyl-1H-pyrazole-4,5-dicarboxylate 45b. Following general procedure 2 using 4-trifluoromethyl-N-benzylsydnone 5e (122 mg, 0.5 mmol) and DMAD (142 mg, 1.0 mmol), dimethyl 1-benzyl-5 trifluoromethyl-1H-pyrazole-3,4-dicarboxylate 45a and dimethyl 1 benzyl-3-trifluoromethyl-1H-pyrazole-4,5-dicarboxylate 45b were isolated as an inseparable mixture $(63:37, 45a/45b)$ as a yellow oil (87) mg, 51%): ¹H NMR (400 MHz, CDCl₃) δ 7.38−7.33 (m, 3H), 7.29− 7.24 (m, 2H), 5.67 (s, 0.75H), 5.54 (s, 1.25H), 3.97 (s, 1.88H), 3.95 (s, 1.88H), 3.90 (s, 1.22H), 3.88 (s, 1.22H); 13C NMR (100.6 MHz, CDCl₃) δ 162.6, 161.6, 160.5, 158.9, 140.6, 139.7 (q, J = 39.0 Hz), 134.7, 134.4, 134.0, 131.0 (q, $J = 40.0$ Hz), 128.9 (2), 128.7, 128.6, 128.0, 127.5, 120.1 (q, $J = 270.0$ Hz), 119.1, 118.8 (q, $J = 271.0$ Hz), 56.4, 56.1, 53.2, 53.1, 52.8 (2), unable to unequivocally assign $1 \times C$ signal; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -58.4 (0.63F), -61.7

(0.37F); FTIR 3029 (w), 2968 (w), 1734 (s), 1496 (m), 1244 (s), 1187 (s), 1134 (s), 1060 (s), 964 (m), 876 (m), 719 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₅H₁₄F₃N₂O₄ 343.0906, found 343.0922.

General Procedure 3: Cycloaddition Reactions of 4- Trifluoromethylsydnones and Alkynylboronates (See Table 2). A solution of a 4-trifluoromethylsydnone (0.5 mmol) and an alkynylboronate (2.0 mmol) in o-DCB (1 M) in a sealed microwave vessel was heated at 140 °C for the designated length of time. After [co](#page-2-0)oling, the crude materials were purified by flash chromatography on silica gel (eluting with 5% EtOAc in petroleum ether, unless otherwise stated) to yield the title pyrazoles.

Synthesis of 1-Phenyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 30a and 1-Phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 30b. Following general procedure 3 using 4 trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 2 ethynyl-4,4,5,5-tetramethyl-1,3,2-dioaborolane (304 mg, 2.0 mmol), heating for 72 h, eluting with 40% EtOAc in petroleum ether, 1 phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 30a and 1-phenyl-3-(4,4,5,5-tetramethyl-1,3,2 dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 30b were isolated as an inseparable mixture (7:93, 30a/30b) as a brown solid (117 mg, 69%): mp 111−113 °C Spectroscopic data only reported for 30b: ¹H NMR (400 MHz, CDCl₃) δ 7.50−7.47 (m, 5H), 7.21 (s, 1H), 1.39 (s, 12H); ¹³C NMR (100.6 MHz, CDCl₃) δ 139.1, 133.2 (q, J = 39.0 Hz), 129.4, 128.8, 126.1, 120.0 (q, $J = 269.0$ Hz), 116.0, 84.5, 24.8; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.5; FTIR 2988 (w), 1561 (w), 1485 (m), 1366 (m), 1290 (m), 1171 (s), 1125 (s), 995 (m), 853 (m), 777 (m), 697 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₆H ₁₉¹¹BF₃N₂O₂ 339.1492, found 339.1477.

Synthesis of 1-Methyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 31b. Following general procedure 3 using 4-trifluoromethyl-N-methylsydnone 5d (84 mg, 0.5 mmol) and 2-ethynyl-4,4,5,5-tetramethyl-1,3,2-dioaborolane (304 mg, 2.0 mmol), heating for 72 h, eluting with 40% EtOAc in petroleum ether, 1-methyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 31b was isolated as a brown oil (122 mg, 44%): ¹H NMR (400 MHz, CDCl₃) δ 7.00 (s, 1H), 4.04 (s, 3H), 1.34 (s, 12H); ¹³C NMR (100.6 MHz, CDCl₃) δ 132.3 (q, J = 39.0 Hz), 120.3 (q, J = 268.5 Hz), 115.0, 84.4, 38.2, 24.8; 19F NMR $(376.5 \text{ MHz}, \text{CDCl}_3)$ δ −60.5; FTIR 2977 (m), 1554 (m), 1236 (s), 1161 (s), 1121 (s), 1050 (s), 993 (m), 856 (m), 719 (m), 635 (w) cm⁻¹; HRMS m/ z [MH]⁺ calcd for C₁₁H₁₇¹¹BF₃N₂O₂ 277.1335, found 277.1328.

Synthesis of 1-Phenyl-3-trimethylsilyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 32a and 1-phenyl-4-trimethylsilyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 32b. Following general procedure 3 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and trimethyl((4,4,5,5-tetramethyl-1,3,2-dixaborolan-2 yl)ethynyl)silane (448 mg, 2.0 mmol), heating for 48 h, 1-phenyl-3 trimethylsilyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 32a was isolated as a yellow solid (120 mg, 58%) and 1-phenyl-4-trimethylsilyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 32b was isolated as a yellow solid (13 mg, 6%).

32a: mp 115−117 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.49−7.46 $(m, 5H)$, 1.40 $(s, 12H)$, 0.40 $(s, 9H)$; ¹³C NMR (100.6 MHz, CDCl₃) δ 159.9, 140.3, 137.8 (q, J = 37.5 Hz), 129.7, 129.5, 126.7, 121.3 (q, J = 270.5 Hz), 84.6, 25.7, 0.0; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –54.5; FTIR 2976 (w), 1504 (w), 1338 (m), 1252 (m), 1139 (s) 1048 (m), 848 (s), 765 (m), 693 (m) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for $C_{19}H_{27}^{11}BF_3N_2O_2Si$ 411.1887, found 411.1890.

32b: mp 113−115 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.46−7.43 (m, 5H), 1.39 (s, 12H), 0.40 (s, 9H); ¹³C NMR (100.6 MHz, CDCl₃) δ 139.3, 136.0 (q, J = 37.5 Hz), 128.4, 128.0, 125.6, 124.1, 120.1 (q, J = 270.5 Hz), 83.7, 24.3, 0.0; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –52.5; FTIR 2980 (w), 1499 (w), 1241 (s), 1168 (s), 1139 (s), 988 (w), 847 (m), 772 (w) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₉H₂₇¹¹BF₃N₂O₂Si 411.1887, found 411.1901.

Synthesis of 1-(p-Nitrophenyl)-3-trimethylsilyl-4-(4,4,5,5 tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1Hpyrazole 33a. Following general procedure 3 using 4-trifluoromethyl-N-(p-nitrophenyl)sydnone (136 mg, 0.5 mmol) and trimethyl- ((4,4,5,5-tetramethyl-1,3,2-dixaborolan-2-yl)ethynyl)silane (448 mg, 2.0 mmol), heating for 24 h, 1-(p-nitrophenyl)-3-trimethylsilyl-4- (4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1Hpyrazole 33a was isolated as a yellow solid (125 mg, 55%).

Note: Compound 33a was isolated as an inseparable mixture (9:1) with the protodeboronated pyrazole: $^1\text{H NMR}$ (400 MHz, CDCl₃) δ 8.37 (d, \hat{J} = 9.0 Hz, 2H), 7.69 (d, J = 9.0 Hz, 2H), 1.40 (s, 12H), 0.40 $(s, 9H)$; ¹³C NMR (100.6 MHz, CDCl₃) δ 160.6, 147.5, 144.5, 137.3 $(q, J = 38.0 \text{ Hz})$, 126.4, 124.4, 123.1 $(q, J = 271.0 \text{ Hz})$, 84.5, 25.0, -0.8 ; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –54.0; FTIR 2984 (w), 1525 (m), 1357 (m), 1256 (s), 1135 (s), 1037 (s), 982 (m), 838 (s), 755 (m) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₉H₂₆¹¹BF₃N₃O₄Si 456.1738, found 456.1756.

Synthesis of 1-Phenyl-3-n-butyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 34a. Following general procedure 3 using 4-trifluoromethyl-N-phenylsydnone 5a (115 mg, 0.5 mmol) and 2-(hex-1-yn-1-yl)-4,4,5,5 tetramethyl-1,3,2-dioxaborolane (416 mg, 2.0 mmol), heating for 72 h, 1-phenyl-3-n-butyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)- 5-trifluoromethyl-1H-pyrazole 34a was isolated as a yellow oil (108 mg, 55%): ¹H NMR (400 MHz, CDCl₃) δ 7.48−7.42 (m, 5H), 2.84− 2.80 (m, 2H), 1.73−1.64 (m, 2H), 1.48−1.39 (m, 2H), 1.38 (s, 12H), 0.96 (t, J = 7.5 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 159.2, 139.5, 136.8 (q, J = 38.0 Hz), 129.0, 128.9, 126.0, 120.1 (q, J = 270.5 Hz), 84.0, 32.7, 28.1, 24.7, 22.7, 13.9; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –55.1; FTIR 2959 (m), 2923 (m), 2872 (w), 1510 (m), 1331 (m), 1239 (s), 1176 (s), 4468 (s), 1076 (s), 984 (m), 855 (m), 771 (m) cm $^{-1}$; HRMS m/z [MH] $^+$ calcd for $\rm C_{20}H_{27}^{-11}BF_3N_2O_2$ 395.2118, found 395.2120.

Synthesis of 3-trifluoromethyl-5-phenyl-1H-pyrazole 37 and 1-(4-Methoxy-3-nitrophenyl)-3-phenyl-5-trifluoromethyl-1Hpyrazole 38. To an ice-cooled stirring suspension of 1-(pmethoxyphenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole 19a (80 mg, 0.25 mmol) in a mixture of acetonitrile and water (4:1, 6.25 mL) was added a cooled solution of CAN (1.37 g, 2.50 mmol) in water (5 mL). The reaction was warmed to ambient temperature, stirred for 22 h, and then concentrated in vacuo. Water (10 mL) was added to the residue which was then extracted with DCM $(3 \times 10 \text{ mL})$. The organic fractions were washed with saturated $NAHCO_{3(aq)}$ then dried over MgSO4 before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 15% EtOAc in petroleum ether) to isolate 3-trifluoromethyl-5 phenyl-1H-pyrazole as a yellow solid 37 (24 mg, 47%) and 1-(4 methoxy-3-nitrophenyl)-3-phenyl-5-trifluoromethyl-1H-pyrazole was isolated as a yellow solid 38 (12 mg, 13%).

37: mp 121–123 °C (lit.³⁷ mp 122–123 °C); ¹H NMR (400 MHz, CDCl₃) δ 10.54 (br, 1H), 7.60–7.58 (m, 2H), 7.49–7.43 (m, 3H), 6.74 (s, 1H); ¹³C NMR ([100](#page-15-0).6 MHz, CDCl₃) δ 145.1, 143.7 (q, J = 38.5 Hz), 129.5, 129.3, 127.9, 125.6, 121.1 (q, J = 267.0 Hz), 101.1; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –62.2; HRMS m/z [MH]⁺ calcd for $C_{10}H_8F_3N_2$ 213.0640, found 213.0630.

38: ¹H NMR (400 MHz, CDCl₃) δ 8.15 (d, J = 2.5 Hz, 1H), 7.87– 7.85 (m, 2H), 7.78 (dd, J = 2.5, 9.0 Hz, 1H), 7.49−7.42 (m, 3H), 7.24 $(d, J = 9.0 \text{ Hz}, 1\text{H}), 7.15 \text{ (s, 1H)}, 4.08 \text{ (s, 3H)}; ^{13}\text{C NMR} (100.6 \text{ MHz},$ CDCl₃) δ 153.4, 152.3, 139.2, 134.1 (q, J = 39.5 Hz), 131.4, 131.3, 131.1, 129.0, 128.9, 125.9, 123.5, 119.6 (q, J = 269.5 Hz), 113.9, 106.6, 57.0; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –57.6; HRMS m/z [MH]⁺ calcd for $C_{17}H_{13}F_3N_3O_3$ 364.0909, found 364.0908.

Synthesis of 3-Trifluoromethyl-5-phenyl-1H-pyrazole 37.37 A solution of 1-benzyl-3-phenyl-5-trifluoromethyl-1H-pyrazole 39a and 1-benzyl-3-trifluoromethyl-5-phenyl-1H-pyrazole 39b (71:[29,](#page-15-0) 39a/39b; 76 mg, 0.25 mmol) in methanol (5 mL) and AcOH (5 drops) was flowed through a H-Cube continuous flow hydrogenator (1 mL min⁻¹) with a 20% Pd(OH)₂/C catalyst cartridge at 80 °C using controlled H_2 mode (70 bar) as a continuous loop. Once the reaction was complete by TLC analysis, the system was washed with methanol (10 mL). The reaction was subsequently neutralized with NaHCO₃ then filtered and the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 20% EtOAc in petroleum ether) to yield 3 trifluoromethyl-5-phenyl-1H-pyrazole 37 as a colorless solid (40 mg, 75%).

Synthesis of 1,3-Diphenyl-5-trifluoromethyl-1H-pyrazole $8a.8$ A solution of 1-phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 5b (80 mg, 0.24 mmol), br[om](#page-14-0)obenzene (74 mg, 0.47 mmol), CsF (108 mg, 0.71 mmol), and $PdCl₂(dppf)$ (8 mg, 0.01 mmol) in a mixture of DME and water (1:1, 1 mL) in a sealed microwave vessel was heated at 140 °C for 30 min in a CEM Microwave Explorer Reactor. After cooling, the volatiles were removed in vacuo before the crude material was purified by flash chromatography on silica gel (eluting with 5% EtOAc in petroleum) to yield 1,3-diphenyl-5-trifluoromethyl-1H-pyrazole 8a as a colorless solid (40 mg, 59%).

Synthesis of 1-Phenyl-5-trifluoromethyl-1H-pyrazol-3(2H) one 35.³⁸ To a stirring suspension of 1-phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 5b (68 mg, 0.20 m[mo](#page-15-0)l) and Na_2CO_3 (21 mg, 0.2 mmol) in ethanol (8 mL) was added H₂O₂ (30% w/w) (2 mL) dropwise. After 30 min, the reaction mixture was concentrated in vacuo. The residue was dissolved in water (20 mL) and then extracted with EtOAc (3×10 mL), and the organic fractions were dried over $MgSO₄$ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 30% EtOAc in petroleum ether) to yield 1-phenyl-5-trifluoromethyl-1H-pyrazol-3(2H)-one 35 as a colorless solid (27 mg, 59%): mp 159−160 °C; ¹ H NMR (400 MHz, CDCl₃) δ 11.26 (br, 1H), 7.55–7.46 (m, 5H), 6.18 (s, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 161.7, 138.0, 133.7 (q, J = 39.5 Hz), 129.4, 129.3, 125.7, 119.1 (q, $J = 270.0$ Hz), 95.1; ¹⁹F NMR (376.5) MHz, CDCl₃) δ –58.3; FTIR 3065 (w), 1592 (m), 1481 (m), 1197 (s), 1120 (s), 1086 (s), 976 (m), 768 (s), 674 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₀H₈F₃N₂O 229.0589, found 229.0597.

Synthesis of 3-Butyl-1,4-diphenyl-5-trifluoromethyl-1H-pyrazole 16a. To a stirring suspension of 3-butyl-1-phenyl-4-(4,4,5,5 tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 34a (79 mg, 0.20 mmol), $PdCl_2(PPh_3)_2$ (16 mg, 0.01 mmol), and K_3PO_4 (1.27 g, 0.60 mmol) in 1,4-dioxane (1 mL) was added bromobenzene (42 μ L, 0.4 mmol) under a nitrogen atmosphere, and the mixture was heated at 85 °C for 24 h. The reaction was quenched with 1 M HCl_(aq) and extracted with DCM (3 \times 10 mL), and the organic fractions were dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 5% EtOAc in petroleum ether) to yield 1,4-diphenyl-4-n-butyl-5-trifluoromethyl-1H-pyrazole 16a as a yellow oil (52 mg, 76%): ¹H NMR (400 MHz, CDCl₃) δ 7.57−7.42 (m, 8H), 7.39−7.37 (m, 2H), 2.66−2.62 (m, 2H), 1.64− 1.56 (m, 2H), 1.37−1.28 (m, 2H), 0.86 (t, J = 7.0 Hz, 3H); 13C NMR (100.6 MHz, CDCl3) δ 152.5, 139.8, 131.2, 130.0, 129.0 (2), 128.2, 127.8, 126.0, 125.6, 123.9, 120.2 (q, J = 270.5 Hz), 31.3, 26.0, 22.5, 13.8; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –55.3; FTIR 3075 (w), 2930 (w), 2861 (w), 1501 (m), 1166 (s), 1119 (s), 990 (m), 755 (m), 694 (s) cm⁻¹; HRMS m / z [MH]⁺ calcd for $C_{20}H_{20}F_3N_2$ 345.1579, found

345.1589.
Synthesis of 1-Phenyl-5-trifluoromethyl-3-trimethylsilyl-1Hpyrazol-4-ol 36. To a stirring suspension of 1-phenyl-3-trimethylsilyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-5-trifluoromethyl-1H-pyrazole 32a (82 mg, 0.20 mmol) and Na_2CO_3 (21 mg, 0.2 mmol) in ethanol (8 mL) was added H_2O_2 (30% w/w) (2 mL) dropwise. After 2 h, the reaction mixture was concentrated in vacuo. The residue was dissolved in water (20 mL) and then extracted with EtOAc (3 \times 10 mL), and the organic fractions were dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 5% EtOAc in petroleum ether) to yield 1-phenyl-5-trifluoromethyl-3-trimethylsilyl-¹H-pyrazol-4-ol ³⁶ as a yellow solid (48 mg, 80%): mp 114−¹¹⁶ °C; ¹ ¹H NMR (400 MHz, CDCl₃) δ 7.48–7.45 (m, 5H), 5.00 (br, 1H), 0.40 (s, 9H); ¹³C NMR (100.6 MHz, CDCl₃) δ 148.4, 142.9, 139.8,

129.0, 128.8, 125.6, 121.3 (q, $J = 268.0$ Hz), 116.1 (q, $J = 37.0$ Hz), -1.4 ; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -55.9; FTIR 3268 (br), 2959 (w), 1898 (w), 1560 (m), 1249 (m), 1110 (s). 976 (m), 833 (s), 768 (m), 690 (m) cm⁻¹; HRMS m/ z [MH]⁺ calcd for C₁₃H₁₆F₃N₂OSi 301.0984, found 301.0972.

Synthesis of 4-(Ethoxycarbonyl)-N-phenylsydnone 52. To a stirring solution of the N-phenylsydnone 51 (4.86 g, 30 mmol) in THF (150 mL) was added n-BuLi in hexanes (36 mmol) dropwise under a nitrogen atmosphere at −78 °C. After 30 min, the solution was transferred via cannula to a stirring solution of ethyl chloroformate (14.28 mL, 150 mmol) in THF (50 mL) at −78 °C and slowly warmed to ambient temperature overnight. The reaction was subsequently quenched with 1 M $\text{HCl}_{(aq)}$ and water (100 mL) and extracted with DCM (3×25 mL), and the organic fractions were dried over $MgSO₄$ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 40% EtOAc in petroleum ether) to yield 4- (ethoxycarbonyl)-N-phenylsydnone 52 as a yellow solid (4.43 g, 63%): mp 105–106 °C (lit.³⁹ mp 100–102 °C); ¹H NMR (400 MHz, CDCl3) δ 7.75−7.70 (m, 1H), 7.66−7.62 (m, 2H), 7.57−7.54 (m, [2H](#page-15-0)), 4.28 (q, $J = 7.0$ Hz, 2H), 1.28 (t, $J = 7.0$ Hz, 3H); ¹³C NMR $(100.6 \text{ MHz}, \text{CDCl}_3)$ δ 164.5, 157.1, 134.8, 132.5, 129.5, 125.0, 99.7, 61.7, 14.1; FTIR 3064 (w), 2994 (w). 1799 (s), 1712 (s), 1447 (m), 1186 (s), 1064 (s), 760 (s), 669 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{11}H_{11}N_2O_4$ 235.0719, found 235.0709.

Synthesis of 4-Hydroxymethyl-N-phenylsydnone 53. To an ice-cooled stirring suspension of LiBH₄ (909 mg, 41.74 mmol) in THF (40 mL) was added 4-(ethoxycarbonyl)-N-phenylsydnone 52 (3.26 g, 13.91 mmol) in THF (40 mL) dropwise under a nitrogen atmosphere, and the mixture was warmed to ambient temperature overnight. The reaction was subsequently quenched with saturated $NAHCO_{3(aq)}$ and then extracted with EtOAc $(3 \times 25 \text{ mL})$, and the organic fractions were dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by trituration using pentanes and ethanol (5:1) to yield 4-hydroxymethyl-N-phenylsydnone 53 as a yellow solid (2.27 g, 83%).

Note: Compound 53 could alternatively be purified by flash chromatography on silica gel (eluting with 80% EtOAc in petroleum ether): mp 98–100 °C (lit.^{30a`} mp 95–97 °C); ¹H NMR (400 MHz, CDCl3) δ 7.86−7.84 (m, 2H), 7.74−7.64 (m, 3H), 4.50 (s, 2H), 4.01 $(br, 1H);$ ¹³C NMR (100.6 [MH](#page-15-0)z, CDCl₃) δ 166.1, 131.0, 129.6, 127.3, 121.5, 105.3, 48.3; FTIR 3351 (br), 3060 (w), 1705 (s), 1479 (m), 1245 (m), 1011 (s), 895 (m), 770 (s), 695 (s) cm⁻¹; HRMS m/z $[MH]^+$ calcd for $C_9H_9N_2O_3$ 193.0613, found 193.0608.

Synthesis of 4-Formyl-N-phenylsydnone 54. To a stirring suspension of $MnO₂$ (4.78 g, 55.00 mmol) in DCM (100 mL) was added 4-hydroxymethyl-N-phenylsydnone 53 (2.11 g, 11.00 mmol) under a nitrogen atmosphere. After being stirred overnight, the reaction mixture was filtered through Celite before the volatiles were removed in vacuo to yield 4-formyl-N-phenylsydnone 54 as a yellow solid (1.81 g, 86%): mp 115−117[°] °C (lit.^{30b} mp 116−118 °C); ¹H NMR (400 MHz, CDCl₃) δ 9.59 (s, 1H), 7.79−7.74 (m, 1H), 7.70− 7.63 (m, 4H); ¹³C NMR (100.6 MHz, CDCl₃) δ 174.7, 166.4, 133.6, 133.2, 129.9, 124.6, 105.1; FTIR 3057 (w), 1778 (s), 1644 (s), 1470 (s), 1374 (s), 1304 (m), 878 (m), 761 (s), 687 (s) cm[−]¹ ; HRMS m/z $[MH]^{+}$ calcd for $C_9H_7N_2O_3$ 191.0457, found 191.0449.

Synthesis of 4-Monofluoromethyl-N-phenylsydnone 55. To an ice-cooled stirring solution of 4-hydroxymethyl-N-phenylsydnone 53 (961 mg, 5.00 mmol) in THF (25 mL) was added deoxofluor (50% solution in THF) (2.58 mL, 6.0 mmol) dropwise under a nitrogen atmosphere. After 1 h, the reaction mixture was warmed to ambient temperature and stirred for a further 1 h. The reaction was subsequently quenched with saturated NaHCO $_{3(aq)}$ (50 mL) and extracted with DCM (3×10 mL), and the organic fractions were dried over MgSO₄ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 30% EtOAc in petroleum ether) to yield 4 monofluoromethyl-N-phenylsydnone 55 as a yellow solid (780 mg, 80%): mp 89−90 °C; ¹ H NMR (400 MHz, CDCl3) δ 7.80−7.68 (m, 5H), 5.19 (d, J = 50.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 167.6, 133.5, 133.0, 130.4, 124.3, 103.1 (d, $J = 22.5$ Hz), 70.2 (d, $J =$ 170.0 Hz); ¹⁹F NMR (376.5 MHz, CDCl₃) δ –196.7 (t, J = 50.0 Hz); FTIR 3069 (w), 1748 (s), 1476 (m), 1269 (m), 1073 (m), 952 (s), 887 (m), 765 (s), 687 (s) cm⁻1; HRMS m/z [MH]⁺ calcd for $C_9H_8FN_2O_2$ 195.0570, found 195.0561.

Synthesis of 4-Difluoromethyl-N-phenylsydnone 56. To an ice-cooled stirring solution of 4-formyl-N-phenylsydnone 54 (1.80 g, 9.47 mmol) in THF (20 mL), in a sealed nalgene bottle, was added deoxofluor (50% solution in THF) (20.32 mL, 94.66 mmol) dropwise under a nitrogen atmosphere. After 1 h, the reaction mixture was warmed to ambient temperature and stirred for a further 47 h. The reaction was subsequently quenched with saturated $NAHCO_{3(aq)}$ (150 mL) and extracted with DCM $(3 \times 25 \text{ mL})$, and the organic fractions were dried over $MgSO₄$ before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 25% EtOAc in petroleum ether) to yield 4 difluoromethyl-N-phenylsydnone 56 as a yellow solid (1.62 g, 81%): mp 91−92 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.78−7.73 (m, 1H), δ 7.71−7.65 (m, 4H), 6.61 (t, J = 52.0 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 165.9, 133.9, 133.1, 130.1, 124.4, 107.3 (t, J = 235.5 Hz), 100.7 (t, J = 27.5 Hz); ¹⁹F NMR (376.5 MHz, CDCl₃) δ -110.7 (d, J $= 52.0$ Hz); FTIR 3071 (w), 2922 (w), 1772 (s), 1478 (m), 1273 (m), 1090 (m), 1014 (s), 979 (s), 907 (m), 792 (s), 985 (s) cm[−]¹ ; HRMS m/z [MH]⁺ calcd for C₉H₇F₂N₂O₂ 213.0476, found 213.0469.

General Procedure 4: Cycloaddition Reactions of 4- Difluoromethyl-N-phenylsydnone and Alkynes (See Table 4). A solution of a 4-difluoromethyl-N-phenylsydnone 56 (106 mg, 0.5 mmol) and an alkyne (2 equiv) in o -DCB (1 mL) in a sealed microwave vessel was heated at 180 °C for 24 h. After cooling, [th](#page-5-0)e crude materials were purified by flash chromatography on silica gel (eluting with 10% EtOAc in petroleum ether unless otherwise stated) to yield the title pyrazoles.

Synthesis of 1,3-Diphenyl-5-difluoromethyl-1H-pyrazole 57a. Following general procedure 4 using ethynylbenzene (102 mg, 1.0 mmol), 1,3-diphenyl-5-difluoromethyl-1H-pyrazole 57a was isolated as a brown oil $(126 \text{ mg}, 93\%)$: ^{1}H NMR $(400 \text{ MHz},$ CDCl3) δ 7.94−7.92 (m, 2H), 7.63−7.60 (m, 2H), 7.58−7.54 (m, 2H), 7.52−7.46 (m, 3H), 7.42−7.38 (m, 1H), 7.10 (s, 1H), 6.70 (t, J = 53.5 Hz, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 152.2, 139.0, 137.6 $(t, J = 30.0 \text{ Hz})$, 132.3, 129.5, 128.9, 128.8, 128.5, 125.9, 125.0, 108.5 (t, J = 235.0, Hz), 104.6; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –110.0 (d, J = −53.5 Hz); FTIR 3082 (w), 1600 (m), 1489 (s), 1437 (m), 1369 (m), 1159 (m), 1074 (s), 1048 (s), 950 (m), 774 (s), 693 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₆H₁₃F₂N₂ 271.1047, found 271.1050.

Synthesis of 2-(1-Phenyl-5-difluoromethyl-1H-pyrazol-3-yl) pyridine 58a and 2-(1-Phenyl-5-difluoromethyl-1H-pyrazol-4 yl)pyridine 58b. Following general procedure 4 using 2-ethynylpyridine (103 mg, 1.0 mmol), eluting with 20% EtOAc in petroleum ether, 2-(1-phenyl-5-difluoromethyl-1H-pyrazol-3-yl)pyridine 58a and 2-(1-phenyl-5-difluoromethyl-1H-pyrazol-4-yl)pyridine 58b isolated as an inseparable mixture (83:17, 58a/58b) as an orange oil (122 mg, 90%): ¹H NMR (400 MHz, CDCl₃) δ 8.70−8.67 (m, 0.83H), 8.66− 8.65 (m, 0.17H), 8.09−8.07 (m, 1H), 7.79−7.75 (m, 1H), 7.72 (t, J = 53.0 Hz, 0.17H), 7.65−7.48 (m, 5.17H), 7.44 (br, 0.83H), 7.30−7.27 (m, 0.83H), 7.25−7.22 (m, 0.17H), 6.69 (t, J = 53.5 Hz, 0.83H); 13C NMR (100.6 MHz, CDCl₃) δ 152.4, 151.1, 150.9, 149.7, 149.6, 149.5, 140.1, 138.9, 137.8 (t, $J = 30.0$ Hz), 136.9, 136.7, 133.0 (t, $J = 24.5$ Hz), 129.5, 129.3, 129.2, 129.0 (x2), 125.7, 125.1, 123.1, 122.0, 121.6, 120.3, 108.7 (t, J = 236.0 Hz), 108.4 (t, J = 236.0 Hz), 106.3; ¹⁹F NMR $(376.5 \text{ MHz}, \text{CDCl}_3) \delta - 110.2 \text{ (d, } J = 53.5 \text{ Hz}, 1.67 \text{ F}), -111.4 \text{ (d, } J =$ 53.0 Hz, 0.33F); FTIR 3050 (w), 1593 (m), 1500 (s), 1415 (m), 1376 (m), 1164 (m), 1075 (s), 1024 (s), 786 (s), 765 (m), 688 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₅H₁₂F₂N₃ 272.0999, found 272.0989.

Synthesis of 1-Phenyl-3-cyclopropyl-5-difluoromethyl-1Hpyrazole 59a. Following general procedure 4 using ethynylcyclopropane (66 mg, 1.0 mmol), 1-phenyl-3-cyclopropyl-5-difluoromethyl-1H-pyrazole 59a was isolated as a yellow oil (89 mg, 76%): ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 7.50–7.49 (m, 4H), 7.45–7.42 (m, 1H), 6.59 (t, $J = 53.5$ Hz, 1H), 6.43 (s, 1H), 2.04 (tt, $J = 5.0$, 8.5 Hz, 1H), 1.03– 0.98 (m, 2H), 0.86–0.82 (m, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 156.1, 139.0, 136.7 (t, $J = 30.0$ Hz), 129.4, 128.5, 124.8, 108.5 (t, $J =$ 235.0 Hz), 103.8, 9.1, 8.2; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -109.8 $(d, J = 53.5 \text{ Hz})$; FTIR 3087 (w), 2991 (w), 1502 (s), 1367 (m), 1154 (m), 1076 (s), 1023 (s), 1005 (s), 800 (m), 754 (s), 690 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{13}H_{13}F_2N_2$ 235.1047, found 235.1041.

Synthesis of 1-Phenyl-3-(3-chloro)propyl-5-difluoromethyl-1H-pyrazole 60a. Following general procedure 4 using 5-chloro-1 pentyne (103 mg, 1.0 mmol), 1-phenyl-3-(3-chloro)propyl-5-difluoromethyl-1H-pyrazole 60a was isolated as a yellow oil (79 mg, 58%): ¹H NMR (400 MHz, CDCl₃) δ 7.20–7.49 (m, 4H), 7.48–7.44 (m, 1H), 6.62 (t, $J = 53.5$ Hz, 1H), 6.61 (s, 1H), 3.65 (t, $J = 6.5$ Hz, 2H), 2.91 (t, J = 7.5 Hz, 2H), 2.25−2.18 (m, 2H); 13C NMR (100.6 MHz, CDCl₃) δ 152.4, 138.9, 136.8 (t, J = 30.0 Hz), 129.5, 128.7, 124.8, 108.5 (t, J = 235.5 Hz), 106.3, 44.3, 32.1, 25.3; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –109.8 (d, J = 53.5 Hz); FTIR 3076 (w), 2956 (w), 1504 (s), 1358 (m), 1158 (m), 1080 (s), 1023 (s), 1001 (s), 816 (m), 767 (s), 687 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₃H₁₄ClF₂N₂ 271.0814, found 271.0809.

Synthesis of Ethyl 1-Phenyl-5-difluoromethyl-1H-pyrazole-3-carboxylate 61a and Ethyl 1-Phenyl-5-difluoromethyl-1Hpyrazole-4-carboxylate 61b.⁴⁰ Following general procedure 4 using ethyl propiolate (98 mg, 1.0 mmol), ethyl 1-phenyl-5 difluoromethyl-1H-pyrazole-3-ca[rbo](#page-15-0)xylate 61a was isolated as a yellow solid (97 mg, 73%) and ethyl 1-phenyl-5-difluoromethyl-1H-pyrazole-4-carboxylate 61b was isolated as a yellow oil (29 mg, 22%).

61a: mp 104−105 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.54 (br, 5H), 7.28 (s, 1H), 6.63 (t, J = 53.5 Hz, 1H), 4.47 (q, J = 7.0 Hz, 2H), 1.43 (t, J = 7.0 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 161.2, 144.5, 138.3, 137.9 (t, J = 30.5 Hz), 129.8, 129.5, 125.5, 109.8, 107.9 $(t, J = 236.5 \text{ Hz})$, 61.5, 14.4; ¹⁹F NMR (376.5 MHz, CDCl₃) δ –110.7 $(d, J = 53.5 \text{ Hz})$; FTIR 3080 (w), 2991 (w), 2942 (w), 1721 (s), 1448 (m), 1390 (s), 1229 (s), 1078 (s), 1039 (s), 999 (s), 755 (m), cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{13}H_{13}F_2N_2O_2$ 267.0945, found 267.0941.

61b: ¹H NMR (400 MHz, CDCl₃) δ 8.11 (s, 1H), 7.57–7.55 (m, 2H), 7.53–7.51 (m, 3H), 7.49 (t, J = 53.0 Hz, 1H), 4.40 (q, J = 7.0 Hz, 2H), 1.42 (t, J = 7.0 Hz, 3H); ¹³C NMR (100.6 MHz, CDCl₃) δ 162.3, 141.9, 139.5, 137.2 (t, J = 24.5 Hz), 129.5, 129.0, 125.7, 116.0, 107.1 (t, J = 237.5 Hz), 61.1, 14.3; ¹⁹F NMR (376.5 MHz, CDCl₃) δ -112.8 $(d, J = 53.0 \text{ Hz})$; FTIR 3074 (w), 2992 (w), 2942 (w), 1705 (s), 1490 (m), 1376 (m), 1233 (s), 1190 (m), 1119 (m), 1029 (s), 975 (m), 779 (s), 689 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for C₁₃H₁₃F₂N₂O₂ 267.0945, found 267.0935.

Synthesis of Dimethyl 5-Hydroxymethyl-1-phenyl-1H-pyrazole-3,4-dicarboxylate 63. A solution of 4-monofluoromethyl-Nphenylsydnone 55 (97 mg, 0.5 mmol) and DMAD (0.12 mL, 1.0 mmol) in o -DCB (1 mL) in a sealed microwave vessel was heated at 100 °C for 24 h. After cooling, the crude material was purified by flash chromatography on silica gel (eluting with 40% EtOAc in petroleum ether) to yield dimethyl 5-hydroxymethyl-1-phenyl-1H-pyrazole-3,4 dicarboxylate ⁶³ as an orange solid (83 mg, 57%): mp 136−¹³⁸ °C; ¹ ¹H NMR (400 MHz, CDCl₃) δ 7.60–7.57 (m, 2H), 7.51–7.48 (m, 3H), 4.76 (s, 2H), 3.97 (s, 3H), 3.85 (s, 3H); 13C NMR (100.6 MHz, CDCl3) δ 162.8, 162.4, 144.0, 141.8, 137.8, 129.5 (2), 125.2, 115.1, 60.5, 52.7, 52.2; FTIR 3399 (br), 2942 (w), 1713 (s), 1473 (m), 1313 (m), 1218 (s), 1104 (s), 988 (m), 762 (s), 693 (s) cm⁻¹; HRMS m/z ${\rm [MH]^+}$ calcd for ${\rm C}_{14}{\rm H}_{15}{\rm N}_2{\rm O}_5$ 291.0981, found 291.0974.

Synthesis of (1,3-Diphenyl-1H-pyrazol-5-yl)methanol 65. A solution of 4-hydroxymethyl-N-phenylsydnone 53 (96 mg, 0.5 mmol) and ethynylbenzene (102 mg, 1.0 mmol) in o-DCB (1 mL) in a sealed microwave vessel was heated at 180 °C for 24 h. After cooling, the crude material was purified by flash chromatography on silica gel (eluting with 40% EtOAc in petroleum ether) to yield (1,3-diphenyl-1H-pyrazol-5-yl)methanol 65 as a yellow solid (180 mg, 72%): mp 111-112 °C (lit.⁴¹ 117-119 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.90−7.88 (m, 2H), 7.68−7.66 (m, 2H), 7.52−7.34 (m, 6H), 6.77 (s, 1H[\),](#page-15-0) 4.69 (s, 2H), 2.19 (br, 1H); ¹³C NMR (100.6 MHz, CDCl₃) δ 151.7, 143.5, 139.5, 132.9, 129.2, 128.7, 128.1, 127.9, 125.8, 124.5, 104.9, 55.7; FTIR 3299 (br), 3068 (w), 2879 (w), 1590 (m), 1502 (s),

1369 (m), 1008 (m), 808 (m), 766 (s), 696 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{16}H_{15}N_2O$ 251.1184, found 251.1175.

Synthesis of 5-Monofluoromethyl-1,3-diphenyl-1H-pyrazole 66 and (1,3-Diphenyl-1H-pyrazol-5-yl)methanol $65.^{41}$ To an icecooled stirring solution of (1,3-diphenyl-1H-pyrazol-5-yl)methanol 65 (100 mg, 0.40 mmol) in THF (1 mL) was added de[oxo](#page-15-0)fluor(50% solution in THF) (0.10 mL, 0.48 mmol) dropwise under a nitrogen atmosphere. After 1 h, the reaction mixture was warmed to ambient temperature and stirred for a further 1 h. The reaction was subsequently quenched with saturated NaHCO_{3(aq)} (5 mL) and extracted with DCM $(3 \times 10 \text{ mL})$, and the organic fractions were dried over MgSO4 before the volatiles were removed in vacuo. The crude material was purified by flash chromatography on silica gel (eluting with 10% EtOAc in petroleum ether) to yield 5 monofluoromethyl-1,3-diphenyl-1H-pyrazole 66 as a yellow solid (32 mg, 32%) and (eluting with 30% EtOAc in petroleum ether) (1,3 diphenyl-1H-pyrazol-5-yl)methanol 65 as a yellow solid (41 mg, 41%).

66: mp 87–89 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.93–7.91 (m, 2H), 7.69−7.67 (m, 2H), 7.57−7.53 (m, 2H), 7.48−7.44 (m, 3H), 7.40−7.35 (m, 1H), 6.95 (d, J = 3.5 Hz, 1H), 5.38 (d, J = 49.0 Hz, 2H); ¹³C NMR (100.6 MHz, CDCl₃) δ 151.9, 139.2, 138.3 (d, J = 19.5 Hz), 132.7, 129.3, 128.7, 128.3, 128.2, 125.8, 124.6, 107.6 (br), 73.8 (d, J = 166.0 Hz); ¹⁹F NMR (376.5 MHz, CDCl₃) δ -196.9 (td, J = 3.5, 49.0 Hz); FTIR 3059 (w), 2975 (w), 1594 (m), 1500 (m), 1434 (m), 1368 (m), 1156 (m), 944 (s), 759 (s), 690 (s) cm⁻¹; HRMS m/z [MH]⁺ calcd for $C_{16}H_{14}N_2F$ 253.1141, found 253.1149.

■ ASSOCIATED CONTENT

S Supporting Information

¹H and ¹³C NMR spectra for selected compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.

■ [AUTHO](http://pubs.acs.org)R INFORMATION

Corresponding Author

*E-mail: j.harrity@sheffield.ac.uk.

Notes

The auth[ors declare no competin](mailto:j.harrity@sheffield.ac.uk)g financial interest.

■ ACKNOWLEDGMENTS

We are grateful to the EPSRC and Bayer CropScience for financial support.

■ REFERENCES

(1) (a) Glasnov, T. N.; Groschner, K.; Kappe, C. O. ChemMedChem 2009, 4, 1816−1818. (b) Lamberth, C. Heterocycles 2007, 71, 1467− 1502.

(2) (a) Singh, S. K.; Saibaba, V.; Rao, K. S.; Reddy, P. G.; Daga, P. R.; Rajjak, S. A.; Misra, P.; Rao, Y. K. Eur. J. Med. Chem. 2005, 40, 977− 990. (b) Cheng, H.; DeMello, K. M. L.; Li, J.; Sakya, S. M.; Ando, K.; Kawamura, K.; Kato, T.; Rafka, R. J.; Jaynes, B. H.; Ziegler, C. B.; Stevens, R.; Lund, L. A.; Mann, D. W.; Kilroy, C.; Haven, M. L.; Nimz, E. L.; Dutra, J. K.; Li, C.; Minich, M. L.; Kolosko, N. L.; Petras, C.; Silvia, A. M.; Seibel, S. B. Bioorg. Med. Chem. Lett. 2006, 16, 2076− 2080.

(3) (a) Lee, H. I.; Le Hir de Fallois, L. P.; Timmons, P. R.; Cawthorne, W. G.; De Leon, A. P. WO 2008005489; Chem. Abstr. 2008, 148, 121701. (b) Jakobi, H.; Ort, O.; Hills, M.; Kehne, H.; Rosinger, C.; Feucht, D. WO 2008080504; Chem. Abstr. 2008, 149, 121249.

(4) (a) Penning, T. D.; Talley, J. J.; Bertenshaw, S. R.; Carter, J. S.; Collins, P. W.; Doctor, S.; Grevato, M. J.; Lee, L. F.; Malecha, J. W.; Miyashiro, J. M.; Rogers, R. S.; Rogier, D. J.; Yu, S. S.; Anderson, G. D.; Burton, E. G.; Gregory, S. A.; Icoboldt, C. M.; Perkus, W. E.; Seibert, K.; Veenhuizen, A. W.; Zhang, Y. Y.; Isakson, P. C. J. Med. Chem. 1997, 40, 1347−1365. (b) Mills, D. L.; Weigel, J. P.; Moyers, T.; Buonomo, F. C. Vet. Ther. 2002, 3, 453−464.

(5) Maxwell, B. D. J. Labelled Compd. Radiopharm. 2000, 43, 645− 654.

(6) Fustero, S.; Sánchez-Roselló, M.; Barrio, P.; Simón-Fuentes, A. Chem. Rev. 2011, 111, 6984−7034.

(7) (a) Song, L.-P.; Zhu, S.-Z. J. Fluorine Chem. 2001, 111, 201−205. (b) Montoya, V.; Pons, J.; Garcia-Anton, J.; Solans, X.; Font-Barcha, M.; Road, J. J. Fluorine Chem. 2007, 128, 1007−1011. (c) Sloop, J. C.; Bumgardner, C. L.; Loehle, W. D. J. Fluorine Chem. 2002, 118, 135− 147. (d) Rüger, A. J.; Nieger, M.; Bräse, S. Tetrahedron 2012, 68, 8823−8829.

(8) Fustero, S.; Román, R.; Sanz-Cervera, J. F.; Simón-Fuentes, A.; Cuñ at, A. C.; Villanova, S.; Murguía, M. J. Org. Chem. 2008, 73, 3523− 3529.

(9) Schlosser, M.; Volle, J.-N.; Leroux, F.; Schenk, K. Eur. J. Org. Chem. 2002, 2913−2920.

(10) Browne, D. L.; Harrity, J. P. A. Tetrahedron 2010, 66, 553−568 and references cited therein.

(11) (a) Huisgen, R.; Grashley, R.; Gotthardt, H.; Schmidt, R. Angew. Chem., Int. Ed. 1962, 1, 48−49. (b) Wu, C.; Fang, Y.; Larock, R. C.; Shi, F. Org. Lett. 2010, 12, 2234−2237. (c) Delauney, T.; Geux, P.; Es-Sayed, M.; Vors, J.-P.; Monteiro, N.; Balme, G. Org. Lett. 2010, 12, 3328−3331. (d) Fang, Y.; Wu, C.; Larock, R. C.; Shi, F. J. Org. Chem. 2011, 76, 8840−8851.

(12) Meazza, G.; Zanardi, G.; Piccardi, P. J. Heterocycl. Chem. 1993, 30, 365−371.

(13) (a) Hanamoto, T.; Suetake, T.; Koga, Y.; Kawanami, T.; Furuno, H.; Inanaga, J. Tetrahedron 2006, 62, 6332−6338. (b) Hanamoto, T.; Suetake, T.; Koga, Y.; Kawanami, T.; Furuno, H.; Inanaga, J. Tetrahedron 2006, 63, 5062−5070.

(14) (a) Browne, D. L.; Helm, M. D.; Plant, A.; Harrity, J. P. A. Angew. Chem., Int. Ed. 2007, 46, 8656−8658. (b) Browne, D. L.; Taylor, J. B.; Plant, A.; Harrity, J. P. A. J. Org. Chem. 2009, 74, 396− 400. (c) Browne, D. L.; Vivat, J. F.; Plant, A.; Gomez-Bengoa, E.; Harrity, J. P. A. J. Am. Chem. Soc. 2009, 131, 7762−7769. (d) Foster, R. S.; Huang, J.; Vivat, J. F.; Browne, D. L.; Harrity, J. P. A. Org. Biomol. Chem. 2009, 7, 4052−4056. (e) Browne, D. L.; Taylor, J. B.; Plant, A.; Harrity, J. P. A. J. Org. Chem. 2010, 75, 984−987. (f) Foster, R. S.; Jakobi, H.; Harrity, J. P. A. Tetrahedron Lett. 2011, 52, 1506− 1508.

(15) For a preliminary account of this work: Foster, R. S.; Jakobi, H.; Harrity, J. P. A. Org. Lett. 2012, 14, 4858−4861.

(16) Soloshonok, V. A.; Gerus, I. I.; Yagupol'skii, Y. L.; Kuhlar, V. P. Zh. Org. Khim. 1987, 23, 2308−2313.

(17) Solohonok, V. A.; Yagupol'skii, Y. L Zh. Org. Khim. 1988, 24, 1638−1644.

(18) (a) Pinhoe Melo, T. M. V. D.; Soares, M. I. L.; d'A Rocha Gonsalives, A. M.; Paixão, J. A.; Beja, A. M.; Silva, M. R.; da Velga, L. A.; Pessoa, J. C. J. Org. Chem. 2002, 67, 4045−4054. (b) Pinhoe Melo, T. M. V. D.; Santos, C. I. A.; d'A Rocha Gonsalives, A. M.; Paixão, J. A.; Beja, A. M.; Silva, M. R. Tetrahedon Lett 2003, 44, 8285−8287. (c) Biron, E.; Kessler, H. J. Org. Chem. 2005, 70, 5183−5189.

(19) Roy, S.; Gregg, B. T.; Gribble, G. W.; Le, V.-D.; Roy, S. Tetrahedron 2011, 67, 2161−2195 and references cited therein.

(20) (a) Kolomeitsex, A. A.; Kadyrov, A. A.; Szczepkowska-Sztolcman, J.; Milewska, M.; Koroniak, H.; Bissky, G.; Barten, J. A.; Röschenthaler, G.-V. Tetrahedron Lett. 2003, 44, 8273-8239. (b) Knauber, T.; Arikan, F.; Rö schenthaler, G.-V.; Gooβen, L. J. Chem.Eur. J. 2011, 17, 2689−2697.

(21) Chem, Q. Y.; Wu, S.-W. J. Chem. Soc., Chem. Commun. 1989, 705−706.

(22) For leading references, see: Chupp, J. P. J. Heterocycl. Chem. 1994, 31, 1377−1380.

(23) Butler, R. N.; Hanniffy, J. M.; Stephens, J. C.; Burke, L. A. J. Org. Chem. 2008, 73, 1354−1364.

(24) Removal of the 4-methoxyphenyl group typically requires 5 equiv of CAN stirring at ambient temperature for 1−4 h; however, the reaction did not progress to full conversion with less than 10 equiv and stirring for 22 h.

(25) Pyrazole 39b was prepared independently by the condensation of benzyl hydrazine with 4,4,4-trifluoromethyl-1-phenylbutane-1,3 dione in 78% yield as a single regioisomer.

(26) Angadiyavar, C. S.; George, M. V. J. Org. Chem. 1971, 36, 1589− 1594.

(27) Tanaka, K.; Maeno, S.; Mitsuhashi, K. Chem. Lett. 1982, 543− 546.

(28) Hydrogenolysis of 39a,b was also carried out on a small scale in a conventional flask using 20% w/w $Pd(OH)_{2}/C$ catalyst to provide 37 in 85% yield according to the method of Barawkar et al.: Barawkar, D. A.; Meru, A.; Bandyopadhyay, A.; Banerjee, A.; Deshpande, A. M.; Athare, C.; Koduru, C.; Khose, G.; Gundu, J.; Mahajan, K.; Patil, P.; Kandalkar, S. R.; Niranjan, S.; Bhosale, S.; De, S.; Mukhopadhyay, S.; Chaudhary, S.; Koul, S.; Singh, U.; Chugh, A.; Palle, V. P.; Mookhtiar, K. A.; Vacca, J.; Chakravarty, P. K.; Nargund, R. P.; Wright, S. D.; Roy, S.; Graziano, M. P.; Singh, S. B.; Cully, D.; Cai, T.-Q. ACS Med. Chem. Lett. 2011, 2, 919−923.

(29) (a) Gosselin, F.; O'Shea, P. D.; Webst, R. A.; Reamer, R. A.; Tillyer, R. D.; Grabowski, E. J. J. Synlett 2006, 3267−3270. (b) Norris, T.; Colon-Cruz, R.; Ripin, D.; David, H. B. Org. Biomol. Chem. 2005, 3, 1844−1849. (c) Li, J.; Lynch, M. P.; DeMello, K. L.; Sakya, S. M.; Cheng, H.; Rafka, R. J.; Bronk, B. S.; James, B. H.; Kilroy, C.; Mann, D. W.; Haven, M. L.; Kolosko, N. L.; Petras, C.; Seibel, S. B.; Lund, L. A. Bioorg. Med. Chem. 2005, 13, 1805−1809.

(30) (a) Tien, H.-J.; Yeh, M.-Y.; Wu, C.-H.; Chen, H.-T. J. Chin. Chem. Soc. 1984, 31, 191−17. (b) Tien, H.-J.; Fang, G.-W.; Lin, S.-T.; Tien, L.-L. J. Chin. Chem. Soc. 1992, 39, 107−110. (c) Tien, L.-L.; Lin, S.-T.; Tien, H.-J.; Fang, G.-W. Can. J. Chem. 1993, 71, 796−800.

(31) The starting material was completely consumed, as judged by TLC analysis, and a single, new spot was evident.

(32) The supplementary crystallographic data for compounds 5a and 56 have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication nos. CCDC 913536 and CCDC 913537, respectively. These data can be obtained free of charge from the Cambridge Crystallographic Data centre via www.ccdc.cam.ac.uk/ data request/cif.

(33) Johsnon, A. W.; Wong, S. C. K. Can. J. Chem. 1966, 44[, 2793](www.ccdc.cam.ac.uk/data_request/cif)− 2803.

[\(34\) Homer, L](www.ccdc.cam.ac.uk/data_request/cif).; Oediger, H. Justus. Liebigs. Ann. Chem. 1959, 627, 142−162.

(35) Huang, G.; Yang, J.; Zhang, X. Chem. Commun. 2011, 47, 5587− 5589.

(36) Soloshonok, V. A.; Yagupol'skii, Y. L.; Kukhar, V. P. Zh. Org. Khim. 1988, 24, 1638−1644.

(37) Dias, H. V. R.; Goh, T. K. H. H. Polyhedron 2004, 23, 273−282. (38) Karanewsky, D. S.; Kalish, V. J.; Robinson, E. D.; Ullman, B. R. WO 2000023421; Chem. Abstr. 2000, 132, 308661.

(39) Zotova, S. A.; Yashunskii, V. G. Zh. Org. Khim. 1965, 1, 2218− 2222.

(40) Okada, I.; Ushie, M.; Nakayama, T.; Fujii, H.; Kakinuma, S. WO 2009116558; Chem. Abstr. 2009, 151, 403297.

(41) Huisgen, R.; Knupfer, H.; Sustmann, R.; Wallbillich, G.; Weberndoerfer, V. Chem. Ber. 1967, 100, 1580−1592.